# Hygroline derivatives from Schizanthus tricolor and their anti-trypanosomatid and antiplasmodial activities 

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#### Abstract

Chemical investigation of the alkaloid extract of the aerial parts of Schizanthus tricolor led to the targeted isolation of 26 hygroline derivatives of which 20 were fully characterized. They have not yet been described in the literature and their structures were established by 1D and 2D NMR, UV and IR spectroscopy, and HRESIMS. The configuration was determined by Gauge-Independent Atomic Orbital NMR chemical shift calculations supported by the advanced statistical method DP4 plus, vibrational circular dichroism, and measurement of optical rotation. Their anti-trypanosomatid, antiplasmodial and cytotoxic activities were measured. Several compounds exhibited low micromolar activity against Plasmodium falciparum. None of the identified molecules was cytotoxic.


## 1. Introduction

The genus Schizanthus Ruiz \& Pav. from the Solanaceae family comprises 12 species and are commonly known as little bird or little butterfly. This genus is primarily native to Chile and the species grow in a large diversity of habitats, from the desert to the coast, as well as from the high Andes to areas cleared of forest in the southern region (Pérez et al., 2006). The genus is characterized by numerous tropane alkaloids. Most of them are ester derivatives from angelic, senecioic, tiglic, itaconic, mesaconic, citraconic or cinnamic acid, which generates numerous positional and configurational isomers. To date over 50 alkaloids have been isolated and characterized from this genus. However, among them only seven pyrrolidine derivatives were described, namely hygrine, hygrolines, cuscohygrines, 4-hydroxyphenylpropanoylhygroline, and 1-methyl-2-(1-methyl-2-pyrrolidinyl)-ethyl-6-deox-y-3-O-angeloyl- $\alpha$-galactopyranoside (Christen et al., 2020). Pyrrolidine derivatives are a source of pharmacologically active lead compounds (Islam and Mubarak, 2020) and the object of recent synthesis due to
"their intriguing biological activities, hallucinogenic characteristics and their utility as pharmacological probes" (Bhat and Tilve, 2011). Therefore, the aim of this study was to investigate the diversity of pyrrolidine derivatives in the species Schizanthus tricolor Grau \& Gronbach (Solanaceae) which is known to be particularly rich in alkaloids. In addition, as part of our ongoing research of undescribed antiprotozoal molecules (Cretton et al., 2010, 2014a, 2014b, 2020), the antiparasitic activity of the isolated hygroline derivatives was evaluated against Trypanosoma brucei rhodesiense (T.b. rhodesiense), Trypanosoma cruzi (T. cruzi), Leishmania donovani (L. donovani), and Plasmodium falciparum (P. falciparum).

## 2. Results and discussion

The alkaloid extract from the aerial parts of S. tricolor was analyzed by LC-HRMS/MS in positive mode. The MS/MS fragment at $m / z$ 144.1390, characteristic of the hygroline moiety was used as a marker to identify putative pyrrolidine alkaloids (see supporting information). An early eluting broad peak displaying ion at $m / z 290[\mathrm{M}+\mathrm{H}]^{+}$, followed

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Fig. 1. LC-HRMS chromatogram of the alkaloid fraction of $S$. tricolor (for sake of clarity $m / z$ values are shown without decimal).
by 17 peaks at $m / z 372[\mathrm{M}+\mathrm{H}]^{+}, 6$ peaks at $m / z 600[\mathrm{M}+\mathrm{H}]^{+}, 4$ peaks at $m / z 420[\mathrm{M}+\mathrm{H}]^{+}$, and 18 peaks at $m / z 454[\mathrm{M}+\mathrm{H}]^{+}$were identified as putative pyrrolidine derivatives (Fig. 1). To isolate the targeted hygroline derivatives, the alkaloid extract was fractionated using flash chromatography. Further purification of the fractions was carried out by semi-preparative HPLC and afforded 20 fully characterized undescribed compounds, 1-20 (Fig. 2).

Compound 1 was obtained as a pale yellow oil, and its molecular formula $\mathrm{C}_{14} \mathrm{H}_{26} \mathrm{NO}_{5}$ was established by HRESIMS data ( $\mathrm{m} / \mathrm{z} 290.1979$ $[\mathrm{M}+\mathrm{H}]^{+}$, calcd for $\mathrm{C}_{14} \mathrm{H}_{27} \mathrm{NO}_{5}, 290.1962$ ). The IR spectrum showed absorption bands attributable to hydroxy ( $3255 \mathrm{~cm}^{-1}$ ) and amine (1591 $\mathrm{cm}^{-1}$ ) groups. Analysis of NMR data (Table 1), and comparison with the literature (Muñoz et al., 1994) indicated the presence of a deoxy hexose sugar unit with proton signals at $\delta_{\mathrm{H}} 4.72,3.49,3.44,3.47,3.86,1.06$ ( $\mathrm{H}-1$ to $\mathrm{H}-6$, respectively) and their corresponding carbon signals at $\delta_{\mathrm{C}}$ $100.1,68.0,70.1,71.5,66.4$ and 16.5. The configuration of the sugar unit was determined by VCD. A positive sign of the intense glycoside band around $1160 \mathrm{~cm}^{-1}$ was observed on the VCD spectrum of 1 (Fig. 3) and indicates an axial glycosidic linkage in $\mathrm{C}-1$, with the ${ }^{1} \mathrm{C}_{4}$ conformation (Taniguchi and Monde, 2007a,b). Therefore, the proton $\mathrm{H}-1$ is oriented in equatorial position and the small coupling constant between $\mathrm{H}-1$ and $\mathrm{H}-2(J=3.6 \mathrm{~Hz})$ indicated that $\mathrm{H}-2$ is in axial position (Fig. 4). Likewise, the coupling constants between H-2 and H-3 ( $J=10.0 \mathrm{~Hz}$ ) imposed an axial orientation for $\mathrm{H}-3$, and an equatorial orientation for $\mathrm{H}-4(J=3.4 \mathrm{~Hz})$. Due to an overlapping of the signals of $\mathrm{H}-2$ to $\mathrm{H}-4$ in DMSO- $d_{6}$, a second measurement was performed in $\mathrm{CD}_{3} \mathrm{OD}$, which made it possible to confirm the spin system (see supporting information, Table S3). From these data, a $\beta$-6-deoxy-L-galactose (also named $\beta$-L-fucose) unit was identified. A hygroline moiety was deduced from eight carbon signals corresponding to two methine C-2' and C-7' ( $\delta_{\mathrm{C}} 64.3$ and 73.5 , respectively) each linked to a heteroatom ( N and O , respectively), four methylenes $\mathrm{C}-3^{\prime}$ to $\mathrm{C}-6^{\prime}$ ( $\delta_{\mathrm{C}} 28.4,21.1,55.8$ and 36.7 , respectively), and two methyl groups $\mathrm{C}-8^{\prime}$ and $\mathrm{C}-9^{\prime}$ ( $\delta_{\mathrm{C}} 21.8$ and 39.2, respectively). A key HMBC correlation between $\mathrm{H}-7^{\prime}$ at $\delta_{\mathrm{H}} 3.78(\mathrm{~m}, 1 \mathrm{H})$ and $\mathrm{C}-1$, as well as a ROE interaction between $\mathrm{H}-7^{\prime}$ and $\mathrm{H}-1$ allowed to connect the hygroline moiety to the anomeric carbon (C-1) of the fucose unit (Fig. 4). Unfortunately, the high flexibility of this molecule impeded the determination of the absolute configuration by comparison of the calculated and experimental VCD spectra. Therefore, the relative configurations at $\mathrm{C}-2^{\prime}$ and $\mathrm{C}-7^{\prime}$ were assigned by calculation of the Smith and Goodman DP4 probability (Smith and Goodman, 2010). Comparison of the experimental and theoretical chemical shifts indicated a $2^{\prime} S^{*} / 7^{\prime} S^{*}$ configuration with $85.9 \%$ probability (see supporting
information). To confirm this result, the absolute configuration of the hygroline moiety was determined by an acidic hydrolysis of 1 followed by the optical rotation measurement of the hygroline moiety $[\alpha]_{D}^{22}-48.2$ (c 0.05, EtOH). A comparison with the literature $[\alpha]_{D}^{24}-46.9$ (c 1.00, EtOH) (Liniger et al., 2013) corroborated the $2^{\prime} S / 7^{\prime} S$ configuration. Likewise, the optical rotation of the resulting sugar was compared to that of L-fucose and confirmed the identity of the 6-deoxy hexose. Compound 1 was thus identified as 1(S)-methyl-2-(1-methyl-2(S)-pyrrolidinyl)ethyl 6-deoxy- $\beta$-galactopyranoside, and named schizanthoside A1.

Compound 2 shares the same HRESIMS protonated molecule [M + $\mathrm{H}]^{+}$at $m / z 290.1979$ (calcd for $\mathrm{C}_{14} \mathrm{H}_{27} \mathrm{NO}_{5}, 290.1962$ ) than 1 , and consequently both compounds are isomers. IR and NMR data are also very similar. To distinguish them, a DP4 calculation was carried out and established a $2^{\prime} R^{*} / 7^{\prime} S^{*}$ configuration at $81.1 \%$ for 2 . An acid hydrolysis of 2 followed by the measurement of the optical rotation of the hygroline moiety $[\alpha]_{D}^{22}+98.2$ (c 0.05, EtOH), and comparison with the literature (Liniger et al., 2013) confirmed the presence of a ( + )-pseudohygroline moiety for 2. L-fucose was also confirmed by comparison of its optical rotation with a standard sample. Compound 2 was thus identified as 1 (S)-methyl-2-(1-methyl-2(R)-pyrrolidinyl)ethyl 6-deoxy- $\beta$-galactopyranoside, and named schizanthoside A2.

The HRMS, IR and NMR data of compound 3 are very similar to those of 1 and 2. Nevertheless, a noticeable difference lies in the multiplicity and the constant coupling of $\mathrm{H}-4$ at $\delta_{\mathrm{H}} 2.77(\mathrm{t}, J=9.2 \mathrm{~Hz}, 1 \mathrm{H})$, indicating that H-4 is axially oriented and characteristic of a 6-deoxy-glucose ring (Fig. 4). According to the DP4 probability, a $2^{\prime} R^{*} / 7^{\prime} S^{*}$ configuration was established ( $82.8 \%$ ). Therefore, compound 3 was identified as $1(S)$ -methyl-2-(1-methyl-2(R)-pyrrolidinyl)ethyl 6-deoxy- $\beta$-glucopyranoside, and named schizanthoside A 3 .

The molecular formula $\mathrm{C}_{19} \mathrm{H}_{33} \mathrm{O}_{6}$ of 4 was established by HRESIMS from the protonated molecule $[\mathrm{M}+\mathrm{H}]^{+}$at $m / z 372.2385$ (calcd for $\mathrm{C}_{19} \mathrm{H}_{34} \mathrm{NO}_{6}, 372.2381$ ). MS/MS fragment at $m / z 83.0499$ (see supporting information) suggested the presence of a monocarboxylic isomeric $\mathrm{C}_{5}$ acid, namely angelic, senecioic or tiglic acids that are known to be present in the genus Schizanthus (Christen et al., 2020). Analysis of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ data revealed signals of two methyl groups at $\delta_{\mathrm{H}} 1.97$ (dq, $J=$ 7.3 and $\left.1.6 \mathrm{~Hz}, \mathrm{H}-4^{\prime \prime}\right)$ and at $\delta_{\mathrm{H}} 1.86\left(\mathrm{t}, J=1.6 \mathrm{~Hz}, \mathrm{H}-5^{\prime \prime}\right)$, and a vinylic proton at $\delta_{\mathrm{H}} 6.18$ (qd, $J=7.3$ and $1.6 \mathrm{~Hz}, \mathrm{H}-3^{\prime \prime}$ ). On the DEPTQ spectrum, a carbonyl signal at $\delta_{\mathrm{C}} 167.1\left(\mathrm{C}-1^{\prime \prime}\right)$, two $\mathrm{sp}^{2}$ carbon signals at $\delta_{\mathrm{C}}$ $127.3\left(\mathrm{C}-2^{\prime \prime}\right)$ and $138.7\left(\mathrm{C}-3^{\prime \prime}\right)$, and two methyl signals at $\delta_{\mathrm{C}} 15.8\left(\mathrm{C}-4^{\prime \prime}\right)$ and $20.5\left(\mathrm{C}-5^{\prime \prime}\right)$ were observed. 2D experiments (COSY and HMBC)






Fig. 2. Chemical structures of compounds 1-20.
allowed to interconnect the different elements, and to identify angelic acid as esterifying acid (De la Fuente et al., 1988). A deshielding of proton $\mathrm{H}-2$ in the fucose ring located the esterification of angelic acid in C-2 (Muñoz et al., 1994). According to the DP4 probability, a $2^{\prime} R * / 7^{\prime} S^{*}$ configuration was established (93.2\%) and the compound was characterized as $1(S)$-methyl-2-(1-methyl-2( $R$ )-pyrrolidinyl)ethyl 6-deoxy-2--$O$-angeloyl- $\beta$-galactopyranoside and named schizanthoside B1.

Compound 5 differs from compound 4 by the nature of the $\mathrm{C}_{5}$ acid esterifying the fucose skeleton in $\mathrm{C}-2$. Indeed, in the ${ }^{1} \mathrm{H}$-NMR spectrum a signal corresponding to the vinylic proton $\mathrm{H}-2^{\prime \prime}$ at $\delta_{\mathrm{H}} 5.82(\mathrm{~d}, J=2.0 \mathrm{~Hz})$ was observed. The COSY experiment demonstrated that $\mathrm{H}-2^{\prime \prime}$ correlated with two methyl groups at $\delta_{\mathrm{H}} 1.96\left(\mathrm{~d}, J=2.0 \mathrm{~Hz}, \mathrm{H}-4^{\prime \prime}\right)$, and at $\delta_{\mathrm{H}} 2.21$ (d, $J=2.0 \mathrm{~Hz}, \mathrm{H}-5^{\prime \prime}$ ). This coupling system and the chemical shifts are characteristic of senecioic acid (De la Fuente et al., 1988). According to the DP4 probability, a $2^{\prime} S^{*} / 7^{\prime} S^{*}$ configuration was established (79.4\%), and 5 was characterized as 1(S)-methyl-2-(1-methyl-2(S)-pyrrolidinyl) ethyl 6-deoxy-2-O-senecioyl- $\beta$-galactopyranoside, and named schizanthoside B2.

A comparison of the NMR data of $\mathbf{6}$ with the literature (Muñoz et al.,
1994) indicated that 6 was characterized by the following structure 1-methyl-2-(1-methyl-2-pyrrolidinyl)ethyl 6-deoxy-3-O-angeloyl-galacto pyranoside. DP4 calculation allowed to attribute a $2^{\prime} S^{*} / 7^{\prime} S^{*}$ configuration and the alkaloid was identified as 1(S)-methyl-2-(1-methyl-2(S)-pyrrolidinyl)ethyl 6-deoxy-3-O-angeloyl- $\beta$-galactopyranoside, and named schizanthoside B3. Compound 6 has an optical rotation value of $[\alpha]_{D}^{22}-71.6$ (c 0.05, MeOH), which differs from the published data $[\alpha]_{D}^{20}+14.2$ (c 1.03, EtOH ) indicating a difference in the stereochemistry of the hygroline and/or the sugar moiety. Unfortunately, no data about the stereochemistry was published, and prevents a comparison of the two stereoisomers.

Compound 7 is similar to 6 except that senecioic acid, characterized by $\mathrm{H}-2^{\prime \prime}$ at $\delta_{\mathrm{H}} 5.72$ and two methyl groups at $\delta_{\mathrm{H}} 1.88\left(\mathrm{H}-4^{\prime \prime}\right)$, and at $\delta_{\mathrm{H}}$ 2.10 ( $\mathrm{H}-5^{\prime \prime}$ ), is esterified in C-3 instead of angelic acid for 6. DP4 calculation allowed to attribute at $80.5 \%$ a $2^{\prime} S^{*} / 7^{\prime} S^{*}$ configuration and 7 was identified as $1(S)$-methyl-2-(1-methyl-2(S)-pyrrolidinyl)ethyl 6-deoxy-3-O-senecioyl- $\beta$-galactopyranoside, and named schizanthoside B4.

Compound 8 is composed of a fucose unit esterified by angelic acid with the characteristic vinylic proton at $\delta_{\mathrm{H}} 6.10$ (qd, $J=7.2$ and 1.7 Hz ,

Table 1
${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data of compounds $\mathbf{1 - 3}$ (600 and 150 MHz , in DMSO- $d_{6}, \delta$ in ppm).

| Position | $\begin{aligned} & 1 \\ & \delta_{\mathrm{H}},(J \text { in } \\ & \mathrm{Hz}) \end{aligned}$ | $\begin{aligned} & \delta_{\mathrm{c}} \\ & \text { type } \end{aligned}$ | $\begin{aligned} & 2 \\ & \delta_{\mathrm{H}},(J \text { in } \\ & \mathrm{Hz}) \end{aligned}$ | $\delta_{\mathrm{C}}$, type | $\begin{aligned} & 3 \\ & \delta_{\mathrm{H}},(J \text { in } \\ & \mathrm{Hz}) \end{aligned}$ | $\delta_{\mathrm{C}}$, type |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | $\begin{aligned} & 4.72, \\ & \text { d (3.6) } \end{aligned}$ | $\begin{aligned} & 100.1, \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & 4.65 \\ & \mathrm{~d}(2.8) \end{aligned}$ | $\begin{aligned} & 100.2, \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & 4.64, \\ & \mathrm{~d}(3.9) \end{aligned}$ | $\begin{aligned} & 99.9, \\ & \text { CH } \end{aligned}$ |
| 2 | $\begin{aligned} & 3.49, \mathrm{dd} \\ & (10.0,3.6) \end{aligned}$ | $\begin{aligned} & 68.0 \\ & \mathrm{CH} \end{aligned}$ | 3.51, m | $\begin{aligned} & 68.3, \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & 3.17, \mathrm{dd} \\ & (9.7,3.9) \end{aligned}$ | $\begin{aligned} & 72.3, \\ & \text { CH } \end{aligned}$ |
| 3 | $\begin{aligned} & 3.44 \\ & \mathrm{~d}(10.0 \\ & 3.4) \end{aligned}$ | $\begin{aligned} & 70.1, \\ & \mathrm{CH} \end{aligned}$ | 3.51, m | $\begin{aligned} & 69.6, \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & 3.33, \mathrm{t} \\ & (9.2) \end{aligned}$ | $\begin{aligned} & 72.9 \\ & \mathrm{CH} \end{aligned}$ |
| 4 | $\begin{aligned} & 3.47, \\ & \text { d (3.4) } \end{aligned}$ | $\begin{aligned} & 71.5 \\ & \text { CH } \end{aligned}$ | 3.47, brs | $\begin{aligned} & 71.7 \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & 2.77, \mathrm{t} \\ & (9.2) \end{aligned}$ | $\begin{aligned} & 75.8, \\ & \text { CH } \end{aligned}$ |
| 5 | $\begin{aligned} & 3.86, q \\ & (6.5) \end{aligned}$ | $\begin{aligned} & 66.4, \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & 3.85, \mathrm{qd} \\ & (6.6,1.4) \end{aligned}$ | $\begin{aligned} & 66.1, \\ & \text { CH } \end{aligned}$ | 3.55, m | $\begin{aligned} & 67.5, \\ & \text { CH } \end{aligned}$ |
| 6 | $\begin{aligned} & 1.06, \\ & \mathrm{~d}(6.5) \end{aligned}$ | $\begin{aligned} & 16.5, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 1.06, \\ & \mathrm{~d}(6.6) \end{aligned}$ | $\begin{aligned} & 16.5 \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 1.09, \\ & \mathrm{~d}(6.3) \end{aligned}$ | $\begin{aligned} & 17.8, \\ & \mathrm{CH}_{3} \end{aligned}$ |
| $2^{\prime}$ | 2.52, m | $\begin{aligned} & 64.3, \\ & \text { CH } \end{aligned}$ | 2.42, m | $\begin{aligned} & 62.5, \\ & \text { CH } \end{aligned}$ | 2.59, m | $\begin{aligned} & 62.7, \\ & \text { CH } \end{aligned}$ |
| $3^{\prime}$ | $\begin{aligned} & 1.93, \mathrm{~m} \\ & 1.62, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 28.4 \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 1.95, \mathrm{~m} \\ & 1.42, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 30.2 \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 2.00, \mathrm{~m} \\ & 1.46, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 29.9 \\ & \mathrm{CH}_{2} \end{aligned}$ |
| $4^{\prime}$ | 1.71, m | $\begin{aligned} & 21.1, \\ & \mathrm{CH}_{2} \end{aligned}$ | 1.62, m | $\begin{aligned} & 21.4, \\ & \mathrm{CH}_{2} \end{aligned}$ | 1.66, m | $\begin{aligned} & 21.3 \\ & \mathrm{CH}_{2} \end{aligned}$ |
| $5^{\prime}$ | $\begin{aligned} & 3.09 \text {, brs } \\ & 2.33 \text {, brs } \end{aligned}$ | $\begin{aligned} & 55.8 \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 2.96, \text { ddd } \\ & \text { (9.5, 7.4, } \\ & 3.4) \\ & 2.15, \\ & d(9.5) \end{aligned}$ | $\begin{aligned} & 56.4 \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 3.04, \mathrm{~m} \\ & 2.26, \\ & \mathrm{~d}(9.2) \end{aligned}$ | $\begin{aligned} & 56.0 \\ & \mathrm{CH}_{2} \end{aligned}$ |
| $6^{\prime}$ | $\begin{aligned} & 1.73, \mathrm{~m} \\ & 1.62, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 36.7 \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 1.83, \text { ddd } \\ & (13.2, \\ & 9.0,3.9) \\ & 1.22, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & \text { 40.6, } \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 1.86, \text { ddd } \\ & (13.4, \\ & 9.4,3.9) \\ & 1.30, \mathrm{td} \\ & (9.4,4.6) \end{aligned}$ | $\begin{aligned} & 40.0 \\ & \mathrm{CH}_{2} \end{aligned}$ |
| $7{ }^{\prime}$ | 3.78, m | $\begin{aligned} & 73.5, \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & 3.61 \text {, ddd } \\ & \text { (9.4, 6.2, } \\ & 3.4 \text { ) } \end{aligned}$ | $\begin{aligned} & 74.1, \\ & \text { CH } \end{aligned}$ | 3.63, m | $\begin{aligned} & 74.5, \\ & \text { CH } \end{aligned}$ |
| $8^{\prime}$ | $\begin{aligned} & 1.17, \\ & \mathrm{~d}(6.0) \end{aligned}$ | $\begin{aligned} & 21.8 \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 1.17, \\ & \mathrm{~d}(6.2) \end{aligned}$ | $\begin{aligned} & 22.5 \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 1.18 \\ & \mathrm{~d}(6.2) \end{aligned}$ | $\begin{aligned} & 22.5, \\ & \mathrm{CH}_{3} \end{aligned}$ |
| $9^{\prime}$ | 2.38, s | $\begin{aligned} & 39.2, \\ & \mathrm{CH}_{3} \end{aligned}$ | 2.23, s | $\begin{aligned} & 39.9 \\ & \mathrm{CH}_{3} \end{aligned}$ | 2.30 s | $\begin{aligned} & 39.5, \\ & \mathrm{CH}_{3} \end{aligned}$ |



Fig. 3. Experimental VCD spectra of compounds 1, 16 and 17.
$\mathrm{H}-3^{\prime \prime}$ ), and two methyl groups at $\delta_{\mathrm{H}} 1.93$ (dq, $J=7.2$ and $1.7 \mathrm{~Hz}, \mathrm{H}-4^{\prime \prime}$ ), and at $\delta_{\mathrm{H}} 1.86\left(\mathrm{t}, J=1.7 \mathrm{~Hz}, \mathrm{H}-5^{\prime \prime}\right)$. The esterification site was located in C-4 by the deshielding of proton $\mathrm{H}-4$ at $\delta_{\mathrm{H}} 5.13(\mathrm{~d}, J=3.2 \mathrm{~Hz})$, and a

HMBC correlation between $\mathrm{H}-4$ and $\mathrm{C}-1^{\prime \prime}$ at $\delta_{\mathrm{C}} 167.0$. DP4 calculation allowed to attribute a $2^{\prime} R^{*} / 7^{\prime} S^{*}$ configuration and the compound was identified as 1(S)-methyl-2-(1-methyl-2(R)-pyrrolidinyl)ethyl 6-deoxy-4-O-angeloyl- $\beta$-galactopyranoside, and named schizanthoside B5.

Compound $\mathbf{9}$ has the same structure as $\mathbf{8}$ except that DP4 calculation determined $2^{\prime} S * / 7^{\prime} S^{*}$ configuration for the hygroline moiety. Therefore, 9 was identified as $1(S)$-methyl-2-(1-methyl-2(S)-pyrrolidinyl)ethyl 6-deoxy-4-O-angeloyl- $\beta$-galactopyranoside, and named schizanthoside B6.

Compound 10 differs from 8 and 9 by the presence of senecioic acid esterified in $\mathrm{C}-4$. The nature of the acid was identified by the proton H $2^{\prime \prime}$ at $\delta_{\mathrm{H}} 5.72$ and two methyl groups at $\delta_{\mathrm{H}} 1.89\left(\mathrm{H}-4^{\prime \prime}\right)$, and at $\delta_{\mathrm{H}} 2.10(\mathrm{H}-$ $5^{\prime \prime}$ ). DP4 calculation allowed to determine a $2^{\prime} S^{*} / 7^{\prime} S^{*}$ configuration and 10 was identified as 1(S)-methyl-2-(1-methyl-2(S)-pyrrolidinyl)ethyl 6-deoxy-4-O-senecioyl- $\beta$-galactopyranoside, and named schizanthoside B7.

The molecular formula $\mathrm{C}_{24} \mathrm{H}_{39} \mathrm{NO}_{7}$ of compound 11 was deduced from the protonated molecule $[\mathrm{M}+\mathrm{H}]^{+}$at $m / z 454.2797$ (calcd for $\mathrm{C}_{24} \mathrm{H}_{40} \mathrm{NO}_{7}, 454.2800$ ). A comparison with the molecular formula of the isomers at $m / z 372.2385$ indicated a difference of $\mathrm{C}_{5} \mathrm{H}_{6} \mathrm{O}$, which could correspond to an additional $\mathrm{C}_{5}$ acid esterified on the fucose moiety. This assumption was corroborated by the analysis of the ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum where two vinylic protons at $\delta_{\mathrm{H}} 6.19\left(1 \mathrm{H}, \mathrm{qd}, J=7.3\right.$ and $\left.1.9 \mathrm{~Hz}, \mathrm{H}-3^{\prime \prime}\right)$ and at $\delta_{\mathrm{H}} 6.11\left(1 \mathrm{H}, \mathrm{qd}, J=7.2\right.$ and $\left.1.5 \mathrm{~Hz}, \mathrm{H}-3^{\prime \prime \prime}\right)$, and four methyl groups at $\delta_{\mathrm{H}} 1.87\left(6 \mathrm{H}, \mathrm{m}, \mathrm{H}-4^{\prime \prime}\right.$ and $\left.\mathrm{H}-4^{\prime \prime \prime}\right)$, and at $\delta_{\mathrm{H}} 1.78(6 \mathrm{H}, \mathrm{dt}, J=8.9$ and $1.6 \mathrm{~Hz}, \mathrm{H}-5^{\prime \prime}$ and $\mathrm{H}-5^{\prime \prime \prime}$ ) were observed. The chemical shifts are characteristic of two angelic acids and their location on the sugar was determined in C-2 and C-3 by the deshielding of protons $\mathrm{H}-2$ at $\delta_{\mathrm{H}} 5.10$ $\left(1 \mathrm{H}, \mathrm{dd}, J=11.0\right.$ and 3.8 Hz ), and $\mathrm{H}-3$ at $\delta_{\mathrm{H}} 5.19(1 \mathrm{H}, \mathrm{dd}, J=11.0$ and 3.2 Hz ). In addition, HMBC correlations were observed from $\mathrm{H}-2$ to $\mathrm{C}-1^{\prime \prime}$ at $\delta_{\mathrm{C}} 166.5$ and from $\mathrm{H}-3$ to $\mathrm{C}-1^{\prime \prime \prime}$ at $\delta_{\mathrm{C}} 166.6$. DP4 calculation determined a $2^{\prime} R^{*} / 7^{\prime} S^{*}$ configuration for the hygroline part and 11 was characterized as 1(S)-methyl-2-(1-methyl-2(R)-pyrrolidinyl)ethyl 6-deoxy-2,3-O-diangeloyl- $\beta$-galactopyranoside, and named schizanthoside C1.

Compound 12 is an isomer of 11 and showed close structural similarities. However, according to NMR data the second angelic acid is esterified in $\mathrm{C}-4$ instead of $\mathrm{C}-3$ in $\mathbf{1 1}$. Indeed, HMBC correlations were observed between $\mathrm{H}-2$ at $\delta_{\mathrm{H}} 4.76(1 \mathrm{H}, \mathrm{dd}, J=10.6$ and 3.8 Hz$)$ and C- $1^{\prime \prime}$ at $\delta_{\mathrm{C}} 166.8$, and between $\mathrm{H}-4$ at $\delta_{\mathrm{H}} 5.21(1 \mathrm{H}, \mathrm{d}, J=3.6 \mathrm{~Hz})$ and $\mathrm{C}-1^{\prime \prime \prime}$ at $\delta_{\mathrm{C}} 166.9$. DP4 calculation determined a $2^{\prime} R^{*} / 7^{\prime} S^{*}$ configuration for the hygroline part and 12 was characterized as $1(S)$-methyl-2-(1-methyl-2 $(R)$-pyrrolidinyl)ethyl 6-deoxy-2,4-O-diangeloyl- $\beta$-galactopyranoside, and named schizanthoside C2.

The HRMS and spectral data of compound 13 were very similar to those of compound 11. Nevertheless, an analysis of the coupling constants of the sugar protons pointed out a difference in the multiplicity. The coupling constant of $\mathrm{H}-4$ at $\delta_{\mathrm{H}} 3.21(1 \mathrm{H}, \mathrm{t}, J=9.3 \mathrm{~Hz})$ indicated that H-4 is axially oriented. DP4 calculation determined a $2^{\prime} R^{*} / 7^{\prime} S^{*}$ configuration for the hygroline moiety and 13 was characterized as $1(S)$ -methyl-2-(1-methyl-2(R)-pyrrolidinyl)ethyl 6-deoxy-2,3-O-diangeloyl-$\beta$-glucopyranoside, and named schizanthoside C3.

Compound 14 contrasts from the previous isomers (11-13) by the presence of tiglic acid revealed by the vinylic proton at $\delta_{\mathrm{H}} 6.67(1 \mathrm{H}, \mathrm{qd}$, $J=7.2$ and $1.8 \mathrm{~Hz}, \mathrm{H}-3^{\prime \prime}$ ) (De la Fuente et al., 1988). HMBC correlation from $\mathrm{H}-3$ at $\delta_{\mathrm{H}} 5.06(1 \mathrm{H}, \mathrm{dd}, J=10.6$ and 3.5 Hz$)$ to $\mathrm{C}-1^{\prime \prime}$ at $\delta_{\mathrm{C}} 166.1$ allowed to locate tiglic acid in C-3. Likewise, a correlation from $\mathrm{H}-4$ at $\delta_{\mathrm{H}}$ $5.24(1 \mathrm{H}, \mathrm{d}, J=3.5 \mathrm{~Hz})$ to $\mathrm{C}-1^{\prime \prime \prime}$ at $\delta_{\mathrm{C}} 166.3$ positioned angelic acid in C-4. A $2^{\prime} S^{*} / 7^{\prime} S^{*}$ configuration was determined by DP4 calculation for the hygroline moiety and 14 was characterized as $1(S)$-methyl-2-(1--methyl-2(S)-pyrrolidinyl)ethyl 6-deoxy-3-O-tigloyl-4-O-angeloyl- $\beta$-glu copyranoside, and named schizanthoside C4.

Compounds 15 and 16 share the same fucose moiety esterified in C-3 and C-4 by two angelic acids. The sites of esterification were elucidated unambiguously by the deshielding of protons $\mathrm{H}-3$ at $\delta_{\mathrm{H}} 5.09(1 \mathrm{H}, \mathrm{dd}, J=$ 10.6 and 3.5 Hz ) for 15 and at $\delta_{\mathrm{H}} 5.11(1 \mathrm{H}, \mathrm{dd}, J=10.5$ and 3.5 Hz$)$ for 16 , and by H-4 at $\delta_{\mathrm{H}} 5.28(1 \mathrm{H}, \mathrm{d}, J=3.5 \mathrm{~Hz})$ for 15 and at $\delta_{\mathrm{H}} 5.26(1 \mathrm{H}$,


Fig. 4. Key HMBC (blue arrows), ROESY (green arrows) correlations and selected coupling constants (red dashed arrow) of compounds 1 and 3. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)


Fig. 5. Key HMBC (blue arrows), ROESY (green arrows) correlations and selected coupling constants (red dashed arrow) of compound 19. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)
dd, $J=3.5$ and 1.3 Hz ) for $\mathbf{1 6}$. For both compounds, HMBC correlations from $\mathrm{H}-3$ to $\mathrm{C}-1^{\prime \prime}$ and $\mathrm{H}-4$ to $\mathrm{C}-1^{\prime \prime \prime}$ confirmed the location of the esterified acids on the fucose moiety. However, a $2^{\prime} S^{*} / 7^{\prime} S^{*}$ configuration for 15 and a $2^{\prime} R^{*} / 7^{\prime} S^{*}$ for 16 were determined by DP4 calculation. Thus, 15 and 16 were described as $1(S)$-methyl-2-(1-methyl-2(S)-pyrrolidinyl) ethyl 6-deoxy-3,4-O-diangeloyl- $\beta$-galactopyranoside and 1(S)-methyl-2-(1-methyl-2(R)-pyrrolidinyl)ethyl 6-deoxy-3,4-O-diangeloyl- $\beta$-galactopyranoside, and named schizanthosides C5 and C6, respectively.

Compound 17 exhibited an HRESIMS protonated molecule $[\mathrm{M}+\mathrm{H}]^{+}$ at $m / z 420.2388$ (calcd for $\mathrm{C}_{23} \mathrm{H}_{34} \mathrm{NO}_{6}, 420.2381$ ), indicating a molecular formula $\mathrm{C}_{23} \mathrm{H}_{33} \mathrm{NO}_{6}$. NMR data showed the presence of a cis-cinnamoyl group characterized by two olefinic methine signals at $\delta_{\mathrm{H}} 6.06$ $\left(1 \mathrm{H}, \mathrm{d}, J=12.8 \mathrm{~Hz}, \mathrm{H}-2^{\prime \prime}\right)$ and at $\delta_{\mathrm{H}} 7.01\left(1 \mathrm{H}, \mathrm{d}, J=12.8 \mathrm{~Hz}, \mathrm{H}-3^{\prime \prime}\right)$, and five aromatic proton signals at $\delta_{\mathrm{H}} 7.35\left(3 \mathrm{H}, \mathrm{m}, \mathrm{H}-6^{\prime \prime}\right.$ to $\left.\mathrm{H}-8^{\prime \prime}\right)$ and $\delta_{\mathrm{H}} 7.66$ ( 2 H , dd, $J=7.4$ and $2.2 \mathrm{~Hz}, \mathrm{H}-5^{\prime \prime}$ and $\mathrm{H}-9^{\prime \prime}$ ). The coupling constant value $J_{2^{\prime \prime}, 3^{\prime \prime}}$ of 12.8 Hz indicated a cis geometry of the double bond. Key HMBC correlation from proton $\mathrm{H}-4$ at $\delta_{\mathrm{H}} 5.11(1 \mathrm{H}, \mathrm{d}, J=3.6 \mathrm{~Hz})$ to C-1" at $\delta_{\mathrm{C}} 165.4$ allowed to determine the connection of the cinnamoyl
moiety to the deoxy galactosyl unit in C-4. A $2^{\prime} R^{*} / 7^{\prime} S^{*}$ configuration was established by DP4 calculation and the compound was identified as 1(S)-methyl-2-(1-methyl-2(R)-pyrrolidinyl)ethyl 6-deoxy-4-O-(Z)-cin-namoyl- $\beta$-galactopyranoside, and named schizanthoside D1.

Compound 18 shares the same protonated molecule $[\mathrm{M}+\mathrm{H}]^{+}$at $m / z$ 420.2388 (calcd for $\mathrm{C}_{23} \mathrm{H}_{34} \mathrm{NO}_{6}, 420.2381$ ) as 17 . As for the latter, a cinnamoyl group was characterized by two olefinic methine signals at $\delta_{\mathrm{H}}$ $6.66\left(1 \mathrm{H}, \mathrm{d}, J=16.0 \mathrm{~Hz}, \mathrm{H}-2^{\prime \prime}\right)$ and $\delta_{\mathrm{H}} 7.65\left(1 \mathrm{H}, \mathrm{d}, J=16.0 \mathrm{~Hz}, \mathrm{H}-3^{\prime \prime}\right)$, and five aromatic proton signals at $\delta_{\mathrm{H}} 7.43\left(3 \mathrm{H}, \mathrm{m}, \mathrm{H}-6^{\prime \prime}\right.$ to $\left.\mathrm{H}-8^{\prime \prime}\right)$ and $\delta_{\mathrm{H}}$ 7.74 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-5^{\prime \prime}$ and $\mathrm{H}-9^{\prime \prime}$ ). Unlike 17 , the coupling constant value $J_{2^{\prime \prime}, 3^{\prime \prime}}$ of 16.0 Hz implied a trans geometry of the double bond. As for 17, a key HMBC correlation from proton $\mathrm{H}-4$ at $\delta_{\mathrm{H}} 5.16(1 \mathrm{H}, \mathrm{d}, J=3.7 \mathrm{~Hz})$ to $\mathrm{C}-1^{\prime \prime}$ at $\delta_{\mathrm{C}} 165.8$ allowed to determine the connection of the cinnamoyl moiety to the 6-deoxygalactopyranosyl ring in C-4. A $2^{\prime} R * / 7^{\prime} S^{*}$ configuration was established by DP4 calculation and the compound was characterized as $1(S)$-methyl-2-(1-methyl-2( $R$ )-pyrrolidinyl)ethyl 6-deoxy-4-O-(E)-cinnamoyl- $\beta$-galactopyranoside, and named schizanthoside D2.

The molecular formula of the alkaloid 19 is $\mathrm{C}_{30} \mathrm{H}_{49} \mathrm{NO}_{11}$, and was deduced from the protonated molecule $[\mathrm{M}+\mathrm{H}]^{+}$at $m / z 600.3386$ (calcd for $\mathrm{C}_{30} \mathrm{H}_{50} \mathrm{NO}_{11}, 600.3379$ ). The analysis of the NMR data showed the presence of a $\beta$-L-fucose unit esterified with a hygroline moiety in C 1 , two tiglic acids in $\mathrm{C}-2$ and $\mathrm{C}-3$, and a rhamnose unit in $\mathrm{C}-4$. The structure of the second sugar unit was inferred from the coupling constant system (1.7 Hz for the anomeric proton, 3.2 Hz between $\mathrm{H}-2^{\prime \prime \prime \prime}$ and $\mathrm{H}-3^{\prime \prime \prime \prime}$, 9.4 Hz between $\mathrm{H}-3^{\prime \prime \prime \prime}, \mathrm{H}-4^{\prime \prime \prime \prime}$ and $\left.\mathrm{H}-5^{\prime \prime \prime \prime}\right)$, as well as a ROESY correlation between $\mathrm{H}-3^{\prime \prime \prime \prime}$ and $\mathrm{H}-5^{\prime \prime \prime \prime}$ (Fig. 5). A $2^{\prime} R^{*} / 7^{\prime} S^{*}$ configuration was determined by DP4 calculation and the compound was identified as 1(S)-methyl-2-(1-methyl-2(R)-pyrrolidinyl)ethyl 6-deoxy-2,3-O-diti-gloyl-4-rhamnosyl- $\beta$-galactopyranoside, and named schizanthoside E1.

Compound 20 is similar to 19 except that two angelic acids are esterified in $\mathrm{C}-2$ and $\mathrm{C}-3$ in the fucose unit instead of two tiglic acids in 19. A $2^{\prime} R * / 7^{\prime} S^{*}$ configuration was determined by DP4 calculation and the compound was identified as $1(S)$-methyl-2-(1-methyl-2 $(R)$-pyrrolidinyl)ethyl 6-deoxy-2,3-O-diangeloyl-4-rhamnosyl- $\beta$-galactopyranoside, and named schizanthoside E2.

Six additional isomers at $m / z 372$ were isolated but not fully characterized due to the low amount of the compounds. Indeed, the NMR spectra did not allow to observe all the carbons and protons of the hygroline moiety (see supporting information, Tables 1 and 2), and consequently to determine the stereochemistry. Nevertheless, the nature of the sugar, the $\mathrm{C}_{5}$ acids and their location on the 6 -deoxy-hexose were elucidated without ambiguity. Their structures are depicted in Fig. 6.

The alkaloid extract and seventeen isolated compounds were tested for their in vitro growth inhibition of T. b. rhodesiense, T. cruzi,

Table 2
${ }^{1} \mathrm{H}$ NMR data of compounds $4-10\left(600 \mathrm{MHz}\right.$, in DMSO- $d_{6}$ or ${ }^{*} \mathrm{CD}_{3} \mathrm{OD}, \delta$ in $\mathrm{ppm}, J$ in Hz ).

| Position | 4 | 5* | 6 | 7 | 8 | 9 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 4.90, d (3.9) | 5.08, d (3.7) | 4.73, d (3.9) | 4.76, d (3.9) | 4.79, d (4.0) | 4.93, d (3.8) | 4.77, d (3.8) |
| 2 | 4.76 , dd (10.6, 3.9) | 4.91, dd (10.5, 3.8) | 3.84 , dd (10.7, 3.9) | 3.70 , dd (10.9, 3.2) | 3.55 , dd (10.3, 3.7) | 3.56 , dd (10.0, 4.0) | 3.44, m |
| 3 | 3.81 , dd (10.5, 3.3) | 3.92 , dd (10.5, 3.4) | 4.83, dd (10.7, 3.2) | 4.80 , dd (10.9, 3.2) | 3.77, dd (10.3, 3.7) | 3.70 , dd (10.0, 3.3) | 3.67 , dd (10.1, 3.6) |
| 4 | 3.56, d (3.4) | 3.73 , d (3.2) | 3.69, d (3.2) | 3.60 , s | 5.13, d (3.2) | $5.14, \mathrm{~d}(3.3)$ | 5.04 , d (3.6) |
| 5 | 3.92, q (6.5) | 4.07, q (6.6) | 3.94, q (6.5) | 3.94, q (6.6) | 4.10, q (6.5) | 4.12, q (6.6) | 4.06, q (6.7) |
| 6 | 1.09, d (6.5) | 1.22, d (6.6) | 1.07, d (6.5) | 1.06, d (6.6) | 0.95, d (6.5) | 0.96, d (6.5) | 0.92, d (6.5) |
| $2^{\prime}$ | 2.10, m | 3.29 | 2.33, m | 2.18, m | 3.10, m | 3.15, m | 2.19, m |
| $3^{\prime}$ | 1.69, m | 2.20, m | 1.93, m | 1.82, m | 2.15, m | 1.94, m | 1.87, m |
|  | 1.30, m | 1.73, m | 1.41, m | 1.54, m | 1.59, m | 1.81, m | 1.51, m |
| $4^{\prime}$ | 1.58, m | 2.00, m | 1.60, m | 1.61, m | 1.81, m | 1.90, m | 1.63, m |
|  | 1.48, m |  |  |  |  | 1.81, m |  |
| $5^{\prime}$ | 2.91, m | 3.61, brs | 2.92, m | 2.87, m | 3.28, m 2.67, m | 3.34, m | 2.95, m |
|  | 1.95, m | 2.99, brs | 2.07, q (8.9) | 2.06, m |  | 2.80, m | 2.09, m |
| $6^{\prime}$ | 1.81, m | 2.13, m | 1.88, m | 1.73, m | 1.92, m | 1.93, m | 1.68, m |
|  | 1.08, m | 1.59, t (12.2) | 1.23, m | 1.56, m | 1.49, m | 1.82, m | 1.59, m |
| $7{ }^{\prime}$ | 3.52, m | 3.70 , m | 3.66, m | 3.78, m | 3.68, m | 3.96, m | 3.74, m |
| $8^{\prime}$ | 1.17, d (6.1) | 1.34, d (6.2) | $1.20, \mathrm{~d}$ (6.3) | 1.18, d (6.1) | 1.21, d (6.2) | 1.22, d (6.1) | 1.16, d (6.2) |
| $9{ }^{\prime}$ | 2.15, s | 2.86, s | 2.20 , s | 2.22, s | 2.52, s | 2.68 , s | 2.24, s |
| $2^{\prime \prime}$ |  | 5.82, d (2.0) |  | 5.72, s |  |  | 5.72, s |
| $3^{\prime \prime}$ | 6.18, qd (7.3, 1.6) |  | 6.08, qd (7.3, 1.6) |  | 6.10, qd (7.2, 1.7) | 6.12, qd (7.2, 1.7) |  |
| $4^{\prime \prime}$ | 1.97, dq $(7.3,1.6)$ | 1.96, d (2.0) | $1.94, \mathrm{dq}(7.3,1.6)$ | 1.88, s | 1.93 , dq (7.2, 1.7) | $1.94, \mathrm{dq}(7.2,1.7)$ | 1.89, s |
| $5^{\prime \prime}$ | 1.86, t (1.6) | 2.21, d (2.0) | 1.86, t (1.6) | 2.10, s | 1.86, t (1.7) | 1.84, t (1.7) | 2.10, s |



6-deoxy-galactose
$21 \mathrm{R}_{1}=\mathrm{H} \mathrm{R}_{2}=\mathrm{H}_{3}=$ =tigloyl
$22 \mathrm{R}_{1}=\mathrm{H} \mathrm{R}_{2}=\mathrm{H} \mathrm{R}_{3}=$ tigloyl



6-deoxy-glucose
$23 \mathrm{R}_{1}=$ angeloyl $\mathrm{R}_{2}=\mathrm{H} \mathrm{R}_{3}=\mathrm{H}$
$24 \mathrm{R}_{1}=\mathrm{H}_{2}=$ angeloyl $\mathrm{R}_{3}=\mathrm{H}$
$25 R_{1}=H R_{2}=H R_{3}=$ angeloyl
$26 \mathrm{R}_{1}=\mathrm{H} \mathrm{R}_{2}=\mathrm{H} \mathrm{R}_{3}=$ angeloyl

Fig. 6. Chemical structures of compounds 21-26.
L. donovani and P. falciparum, which are the etiological agents of sleeping sickness, Chagas disease, visceral leishmaniasis and tropical malaria, respectively. The cytotoxicity in uninfected rat skeletal L6 cells was also evaluated. The data obtained are summarized in Table 7. A significant inhibiton ( $\mathrm{IC}_{50}<10 \mu \mathrm{~g} / \mathrm{mL}$ ) of the alkaloid extract was observed only towards $P$. falciparum. Among the isolated compounds, compounds 11-16 and 18 inhibited the etiologic agent of malaria with $\mathrm{IC}_{50}<10 \mu \mathrm{M}$. Thus the presence of two $\mathrm{C}_{5}$ acids esterifying the sugar unit appears to be essential for the antiplasmodial activity because compounds with one $C_{5}$ acid only (compounds $4,6-10$ ) or without $C_{5}$ acid (compounds 1-3) were inactive against the parasite (Table 7). The nature of the $\mathrm{C}_{5}$ acid (angelic, senecioic or tiglic acid), their location on the sugar unit as well as the stereochemistry of the hygroline seemed to have a low influence on the antiplasmodial activity. In addition, the transcinnamoyl moiety esterified in C-4 for 18 was beneficial for the
antiplasmodial activity ( $\mathrm{IC}_{50}=6.3 \mu \mathrm{M}$ ), whereas the cis-cinnamoyl moiety for 17 was detrimental ( $\mathrm{IC}_{50}=26.1 \mu \mathrm{M}$ ). No cytotoxicity was detected for all compounds.

## 3. Conclusions

The targeted investigation of pyrrolidine derivatives in the aerial parts of S. tricolor resulted in the isolation and characterization of 20 undescribed alkaloids named schizanthosides A1-A3 (1-3), B1-B7 (4-10), C1-C6 (11-16), D1-D2 (17-18), and E1-E2 (19-20). Six other pyrrolidine alkaloids were also isolated but not fully characterized due to lack of material. Compounds 11 - $\mathbf{1 6}$ and 18 showed a significant inhibitory activity towards $P$. falciparum with an $\mathrm{IC}_{50}<10 \mu \mathrm{M}$.

Table 3
${ }^{13} \mathrm{C}$ NMR data of compounds 4-10 (150 MHz, in DMSO- $d_{6}$ or $* \mathrm{CD}_{3} \mathrm{OD}, \delta$ in ppm, C type).

| Position | 4 | 5* | 6 | 7 | 8 | 9 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 97.2, CH | 99.4, CH | 100.0, CH | 100.0, CH | 100.6, CH | 100.1, CH | 99.9, CH |
| 2 | 71.4, CH | 72.4, CH | 65.3, CH | 65.4, CH | 68.8, CH | 68.3, CH | 68.6, СH |
| 3 | 66.9, CH | 69.1, CH | 73.5, CH | 73.0, CH | 67.4, CH | 67.6, CH | 68.0, CH |
| 4 | 71.8, CH | 73.7, CH | 68.9, CH | 69.2, CH | 73.9, CH | 73.8, CH | 73.0, CH |
| 5 | 66.0, CH | 67.9, CH | 65.8, CH | 66.2, CH | 64.8, CH | 65.2, CH | 64.9, CH |
| 6 | 16.3, $\mathrm{CH}_{3}$ | 16.4, $\mathrm{CH}_{3}$ | 16.3, $\mathrm{CH}_{3}$ | 16.3, $\mathrm{CH}_{3}$ | 16.2, $\mathrm{CH}_{3}$ | 16.3, $\mathrm{CH}_{3}$ | 16.2, $\mathrm{CH}_{3}$ |
| $2^{\prime}$ | 61.9, CH | 67.4, CH | 62.2, CH | 63.8, CH | 64.0, CH | 65.5, CH | 63.7, CH |
| $3^{\prime}$ | $30.1 \mathrm{CH}_{2}$ | 30.3, $\mathrm{CH}_{2}$ | $30.4, \mathrm{CH}_{2}$ | 28.7, $\mathrm{CH}_{2}$ | 28.9, $\mathrm{CH}_{2}$ | 27.3, $\mathrm{CH}_{2}$ | 29.1, $\mathrm{CH}_{2}$ |
| $4^{\prime}$ | 21.4, $\mathrm{CH}_{2}$ | 22.2, $\mathrm{CH}_{2}$ | $21.5, \mathrm{CH}_{2}$ | 21.2, $\mathrm{CH}_{2}$ | 20.8, $\mathrm{CH}_{2}$ | 20.7, $\mathrm{CH}_{2}$ | 21.3, $\mathrm{CH}_{2}$ |
| $5^{\prime}$ | 56.6, $\mathrm{CH}_{2}$ | 57.1, $\mathrm{CH}_{2}$ | 56.5, $\mathrm{CH}_{2}$ | 56.2, $\mathrm{CH}_{2}$ | 55.3, $\mathrm{CH}_{2}$ | $55.2, \mathrm{CH}_{2}$ | 56.2, $\mathrm{CH}_{2}$ |
| $6^{\prime}$ | 41.3, $\mathrm{CH}_{2}$ | 39.0, $\mathrm{CH}_{2}$ | 40.7, $\mathrm{CH}_{2}$ | $37.5, \mathrm{CH}_{2}$ | 38.2, $\mathrm{CH}_{2}$ | 34.6, $\mathrm{CH}_{2}$ | 38.0, $\mathrm{CH}_{2}$ |
| $7{ }^{\prime}$ | 74.8, CH | 75.9, CH | 74.1, CH | 73.7, CH | 75.0, CH | 74.1, CH | 73.8, CH |
| $8{ }^{\prime}$ | 22.6, $\mathrm{CH}_{3}$ | 22.5, $\mathrm{CH}_{3}$ | $22.5, \mathrm{CH}_{3}$ | 21.9, $\mathrm{CH}_{3}$ | 22.2, $\mathrm{CH}_{3}$ | 21.7, $\mathrm{CH}_{3}$ | 21.9, $\mathrm{CH}_{3}$ |
| $9^{\prime}$ | 39.9, $\mathrm{CH}_{3}$ | 39.4, $\mathrm{CH}_{3}$ | 40.0, $\mathrm{CH}_{3}$ | 39.6, $\mathrm{CH}_{3}$ | $38.5, \mathrm{CH}_{3}$ | 38.1, $\mathrm{CH}_{3}$ | 39.8, $\mathrm{CH}_{3}$ |
| $1^{\prime \prime}$ | 167.1, C | 167.6, C | 167.3, C | 165.6, C | 167.0, C | 167.0, C | 165.6, C |
| $2^{\prime \prime}$ | 127.3, C | 116.5, CH | 128.0, C | 116.3, CH | 127.8, C | 127.7, C | 116.0, CH |
| $3^{\prime \prime}$ | 138.7, CH | 160.5, C | 136.8, CH | 155.9, C | 136.8, CH | 137.1, CH | 156.3, C |
| $4^{\prime \prime}$ | 15.8, $\mathrm{CH}_{3}$ | 27.5, $\mathrm{CH}_{3}$ | $15.5, \mathrm{CH}_{3}$ | 26.8, $\mathrm{CH}_{3}$ | 15.5, $\mathrm{CH}_{3}$ | $15.6, \mathrm{CH}_{3}$ | 26.9, $\mathrm{CH}_{3}$ |
| $5^{\prime \prime}$ | 20.4, $\mathrm{CH}_{3}$ | $20.5, \mathrm{CH}_{3}$ | 20.3, $\mathrm{CH}_{3}$ | 19.9, $\mathrm{CH}_{3}$ | 20.4, $\mathrm{CH}_{3}$ | 20.5, $\mathrm{CH}_{3}$ | 20.0, $\mathrm{CH}_{3}$ |

Table 4
${ }^{1} \mathrm{H}$ NMR data of compounds $11-16$ ( 600 MHz , in DMSO- $d_{6}, \delta$ in ppm, $J$ in Hz ).

| Position | 11 | 12 | 13 | 14 | 15 | 16 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | $\begin{aligned} & 5.01, \\ & \mathrm{~d}(3.8) \end{aligned}$ | $\begin{aligned} & 5.00, \\ & \mathrm{~d}(3.9) \end{aligned}$ | $\begin{aligned} & 5.03, \\ & \mathrm{~d}(3.9) \end{aligned}$ | $\begin{aligned} & 4.85, \\ & \mathrm{~d}(3.9) \end{aligned}$ | $\begin{aligned} & \text { 4.86, } \\ & \text { d (3.8) } \end{aligned}$ | $\begin{aligned} & \text { 4.91, } \\ & \mathrm{d}(3.9) \end{aligned}$ |
| 2 | $\begin{aligned} & \text { 5.10, dd } \\ & \text { (11.0, } \\ & 3.8) \end{aligned}$ | 4.76, dd (10.6, 3.8) | $\begin{aligned} & \text { 4.64, dd } \\ & \text { (10.5, } \\ & 3.8) \end{aligned}$ | $\begin{aligned} & 3.79, \mathrm{dd} \\ & (10.6, \\ & 3.8) \end{aligned}$ | $\begin{aligned} & 3.80, \mathrm{dd} \\ & \text { (10.6, } \\ & 3.8) \end{aligned}$ | $\begin{aligned} & 3.70, \mathrm{dd} \\ & \text { (10.5, } \\ & 3.8) \end{aligned}$ |
| 3 | 5.19, dd (11.0, 3.2) | 4.09, dd (10.6, 3.6) | 5.31, dd (10.5, 9.3) | 5.06, dd (10.6, 3.5) | 5.09, dd (10.6, 3.5) | $\begin{aligned} & 5.11, \mathrm{dd} \\ & \text { (10.5, } \\ & 3.5) \end{aligned}$ |
| 4 | $\begin{aligned} & 3.80, \\ & \mathrm{~d}(3.6) \end{aligned}$ | $\begin{aligned} & 5.21, \\ & \mathrm{~d}(3.6) \end{aligned}$ | $\begin{aligned} & 3.21, \mathrm{t} \\ & (9.3) \end{aligned}$ | $\begin{aligned} & 5.24, \\ & \mathrm{~d}(3.5) \end{aligned}$ | $\begin{aligned} & 5.28, \\ & \text { d (3.5) } \end{aligned}$ | $\begin{aligned} & 5.26, \mathrm{dd} \\ & (3.5, \\ & 1.3) \end{aligned}$ |
| 5 | $\begin{aligned} & \text { 4.04, dd } \\ & \text { (6.5, } \\ & 1.3 \text { ) } \end{aligned}$ | $\begin{aligned} & 4.17, q \\ & (6.5) \end{aligned}$ | 3.78, m | $\begin{aligned} & 4.21, \mathrm{dd} \\ & \text { (6.4, } \\ & 1.3) \end{aligned}$ | 4.21, m | $\begin{aligned} & 4.23, \mathrm{dd} \\ & \text { (6.4, } \\ & 1.3) \end{aligned}$ |
| 6 | $\begin{aligned} & 1.12, \\ & \mathrm{~d}(6.5) \end{aligned}$ | $\begin{aligned} & 0.98, \\ & \mathrm{~d}(6.5) \end{aligned}$ | $\begin{aligned} & 1.18 \\ & \mathrm{~d}(6.3) \end{aligned}$ | $\begin{aligned} & 1.00, \\ & \mathrm{~d}(6.6) \end{aligned}$ | $\begin{aligned} & 1.00, \\ & \mathrm{~d}(6.8) \end{aligned}$ | $\begin{aligned} & 1.11, \\ & \mathrm{~d}(6.5) \end{aligned}$ |
| $2^{\prime}$ | 2.06, m | 2.08, m | 2.06, m | 2.33, m | $\begin{aligned} & 2.33, \\ & \text { d (9.9) } \end{aligned}$ | 2.15, m |
| $3^{\prime}$ | $\begin{aligned} & 1.68, \mathrm{~m} \\ & 1.30, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 1.69, \mathrm{~m} \\ & 1.29, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 1.66, \mathrm{~m} \\ & 1.29, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 1.95, \mathrm{~m} \\ & 1.40, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 1.95, \mathrm{~m} \\ & 1.42, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 1.83, \mathrm{~m} \\ & 1.54, \mathrm{~m} \end{aligned}$ |
| $4^{\prime}$ | $\begin{aligned} & 1.57, \mathrm{~m} \\ & 1.48, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 1.57, \mathrm{~m} \\ & 1.48, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 1.57, \mathrm{~m} \\ & 1.46, \mathrm{~m} \end{aligned}$ | 1.59, m | 1.60, m | 1.62, m |
| $5^{\prime}$ | $\begin{aligned} & 2.87, \mathrm{~m} \\ & 1.90, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 2.89, \mathrm{~m} \\ & 1.93, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 2.86, \mathrm{~m} \\ & 1.88, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 2.91 \mathrm{~m} \\ & 2.06 \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 2.91 \mathrm{~m} \\ & 2.09 \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 2.85, \mathrm{~m} \\ & 2.01, \\ & \mathrm{~d}(9.0) \end{aligned}$ |
| $6^{\prime}$ | $\begin{aligned} & 1.85, \mathrm{~m} \\ & 1.09, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 1.84, \mathrm{~m} \\ & 1.09, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 1.87, \mathrm{~m} \\ & 1.10, \mathrm{~m} \end{aligned}$ | $1.89, \mathrm{~m}$ | $\begin{aligned} & 1.90, \mathrm{~m} \\ & 1.25, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 1.76, \mathrm{~m} \\ & 1.58, \mathrm{~m} \end{aligned}$ |
| $7{ }^{\prime}$ | 3.59, m | 3.57, m | 3.62, m | 3.71 , m | 3.73, m | 3.82, m |
| $8^{\prime}$ | $\begin{aligned} & 1.22, \\ & \mathrm{~d}(6.1) \end{aligned}$ | $\begin{aligned} & 1.19, \\ & \mathrm{~d}(6.1) \end{aligned}$ | $\begin{aligned} & 1.23 \\ & \mathrm{~d}(6.2) \end{aligned}$ | $\begin{aligned} & 1.22, \\ & \mathrm{~d}(6.2) \end{aligned}$ | $\begin{aligned} & 1.23, \\ & \text { d (6.1) } \end{aligned}$ | $\begin{aligned} & 1.21, \\ & \mathrm{~d}(6.5) \end{aligned}$ |
| $9{ }^{\prime}$ | 2.13, s | 2.14, s | 2.13, s | 2.19, s | 2.20, s | 2.22, s |
| $3^{\prime \prime}$ | $\begin{aligned} & \text { 6.19, qd } \\ & \text { (7.3, } \\ & 1.9) \end{aligned}$ | $\begin{aligned} & 6.20, ~ q d \\ & (7.3, \\ & 1.7) \end{aligned}$ | $\begin{aligned} & 6.21, \mathrm{qd} \\ & \text { (7.3, } \\ & 1.7) \end{aligned}$ | $\begin{aligned} & 6.67, q d \\ & \text { (7.2, } \\ & 1.8) \end{aligned}$ | $\begin{aligned} & 6.08, q \\ & (7.2) \end{aligned}$ | $\begin{aligned} & 6.07, \mathrm{qd} \\ & (7.2, \\ & 1.6) \end{aligned}$ |
| $4 \prime$ | 1.87, m | 1.96, m | 1.89, m | $\begin{aligned} & 1.73, \mathrm{dd} \\ & \text { (7.0, } \\ & 1.4) \end{aligned}$ | 1.87, m | 1.86, m |
| $5^{\prime \prime}$ | $\begin{aligned} & 1.78, \mathrm{dt} \\ & \text { (8.9, } \\ & 1.6) \end{aligned}$ | $\begin{aligned} & 1.87, \mathrm{dt} \\ & \text { (8.9, } \\ & 1.6) \end{aligned}$ | $\begin{aligned} & 1.75, \mathrm{t} \\ & (1.6) \end{aligned}$ | $\begin{aligned} & 1.70, \mathrm{t} \\ & (1.4) \end{aligned}$ | 1.74, m | 1.74, m |
| $3^{\prime \prime \prime}$ | $\begin{aligned} & 6.11, \mathrm{qd} \\ & \text { (7.2, } \\ & 1.5) \end{aligned}$ | 6.14, qd (7.2, 1.5) | $\begin{aligned} & 6.05, \mathrm{qd} \\ & \text { (7.2, } \\ & 1.6) \end{aligned}$ | 6.16, qd (7.2, <br> 1.6) | $\begin{aligned} & 6.17, q \\ & (7.4) \end{aligned}$ | 6.17, qd (7.1, <br> 1.7) |
| $4^{\prime \prime \prime}$ | 1.87, m | 1.96, m | 1.84, m | 1.90, m | $\begin{aligned} & 1.92, \mathrm{dd} \\ & \text { (7.4, } \\ & 1.7) \end{aligned}$ | $\begin{aligned} & 1.92, \\ & \mathrm{~d}(7.3) \end{aligned}$ |
| $5^{\prime \prime \prime}$ | $\begin{aligned} & 1.78, \mathrm{dt} \\ & \text { (8.9, } \\ & 1.6) \end{aligned}$ | $\begin{aligned} & 1.87, \mathrm{dt} \\ & \text { (8.9, } \\ & 1.6) \end{aligned}$ | $\begin{aligned} & 1.79, \mathrm{t} \\ & (1.6) \end{aligned}$ | $\begin{aligned} & 1.86, \mathrm{t} \\ & (1.6) \end{aligned}$ | 1.87, m | 1.86, m |

Table 5
${ }^{13} \mathrm{C}$ NMR data of compounds $11-16$ ( 150 MHz , in DMSO- $d_{6}, \delta$ in ppm, C type).

| Position | 11 | 12 | 13 | 14 | 15 | 16 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 97.1, CH | 97.2, CH | 96.3, CH | 99.5, CH | 99.7, CH | $\begin{aligned} & 100.2, \\ & \text { CH } \end{aligned}$ |
| 2 | 68.3, CH | 71.4, CH | 71.4, CH | 65.9, CH | 66.2, CH | 66.8, CH |
| 3 | 70.1, CH | 64.7, CH | 71.2, CH | 70.3, CH | 70.3, CH | 71.2, CH |
| 4 | 68.9, CH | 74.0, CH | 72.9, CH | 70.6, CH | 71.0, CH | 71.5, CH |
| 5 | 65.9, CH | 64.5, CH | 67.3, СH | 63.8, CH | 64.1, CH | 64.9, CH |
| 6 | $\begin{aligned} & 16.1, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 16.1, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 17.0, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 15.6, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 15.9, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 16.4, \\ & \mathrm{CH}_{3} \end{aligned}$ |
| $2^{\prime}$ | 61.8, CH | 61.8, CH | 61.5, CH | 61.9, CH | 62.2, CH | 64.1, CH |
| $3^{\prime}$ | $30.1 \text {, }$ | $\begin{aligned} & 30.1, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 29.9, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 30.1, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 30.4, \\ & \mathrm{CH}_{2} \end{aligned}$ | $29.2,$ |
| $4^{\prime}$ | $\begin{aligned} & 21.4, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 21.3, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 21.1, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 21.2, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 21.5, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 21.7, \\ & \mathrm{CH}_{2} \end{aligned}$ |
| $5^{\prime}$ | $\begin{aligned} & 56.7, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 56.6, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 56.4, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 56.2, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 56.5, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 56.7, \\ & \mathrm{CH}_{2} \end{aligned}$ |
| $6^{\prime}$ | $\begin{aligned} & 41.4, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 41.3, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 41.3, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 40.6, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 40.7, \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 38.1, \\ & \mathrm{CH}_{2} \end{aligned}$ |
| $7{ }^{\prime}$ | 75.3, CH | 75.3, CH | 75.4, CH | 74.4, CH | 74.6, CH | 74.6, CH |
| $8^{\prime}$ | $\begin{aligned} & 22.7, \\ & \mathrm{CH}_{3} \end{aligned}$ | $22.6,$ | $\begin{aligned} & 22.5 \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 22.1, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 22.4, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 22.3, \\ & \mathrm{CH}_{3} \end{aligned}$ |
| $9{ }^{\prime}$ | $\begin{aligned} & 39.9, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 39.9, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 39.7, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 39.9, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 40.0, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 40.2, \\ & \mathrm{CH}_{3} \end{aligned}$ |
| $1^{\prime \prime}$ | 166.5, C | 166.8, C | 165.9, C | 166.1, C | 166.5, C | 167.0, C |
| $2^{\prime \prime}$ | 127.3, C | 127.1, C | 126.2, C | 127.6, C | 127.3, C | 127.8, C |
| $3^{\prime \prime}$ | $\begin{aligned} & 139.5, \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & \text { 139.3, } \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & 139.9, \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & \text { 136.9, } \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & \text { 137.8, } \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & \text { 137.9, } \\ & \text { CH } \end{aligned}$ |
| $4^{\prime \prime}$ | $\begin{aligned} & 15.4, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 15.6, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 15.4, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 13.9, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 15.4, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 15.8, \\ & \mathrm{CH}_{3} \end{aligned}$ |
| $5^{\prime \prime}$ | $\begin{aligned} & 20.1, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 20.4, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 19.8 \\ & \mathrm{CH}_{2} \end{aligned}$ | $\begin{aligned} & 11.4 \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 19.9 \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 20.4, \\ & \mathrm{CH}_{3} \end{aligned}$ |
| $1^{\prime \prime \prime}$ | 166.6, C | 166.9, C | 166.3, C | 166.3, C | 166.6, C | 167.1, C |
| $2^{\prime \prime \prime}$ | 126.7, C | 127.6, C | 127.2, C | 126.8, C | 127.0, C | 127.5, C |
| $3^{\prime \prime \prime}$ | $\begin{aligned} & \text { 137.8, } \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & \text { 137.3, } \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & 136.4, \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & \text { 137.5, } \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & 138.2, \\ & \text { CH } \end{aligned}$ | $\begin{aligned} & 138.6, \\ & \text { CH } \end{aligned}$ |
| $4^{\prime \prime \prime}$ | $\begin{aligned} & 15.6, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 15.7, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & \text { 15.1, } \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 15.1, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 15.5, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 16.0, \\ & \mathrm{CH}_{3} \end{aligned}$ |
| $5^{\prime \prime \prime}$ | $\begin{aligned} & 20.2, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 20.4, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 20.0, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 20.0, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 20.3, \\ & \mathrm{CH}_{3} \end{aligned}$ | $\begin{aligned} & 20.7, \\ & \mathrm{CH}_{3} \end{aligned}$ |

## 4. Experimental

### 4.1. General experimental procedures

Optical rotations were measured on a JASCO P-1030 (Easton, MD, USA) polarimeter. VCD spectra were recorded on a Bruker PMA 50 accessory coupled to a Tensor 27 Fourier transform infrared spectrometer (Billerica, MA, USA). A photoelastic modulator (Hinds PEM 90,

Table 6
${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data of compounds $\mathbf{1 7 - 2 0}$ ( 600 and 150 MHz , in DMSO- $d_{6}, \delta$ in ppm).

| Position | $\begin{aligned} & 17 \\ & \delta_{\mathrm{H}},(J \text { in } \mathrm{Hz}) \end{aligned}$ | $\delta_{\text {c }}$, type | $\begin{aligned} & 18 \\ & \delta_{\mathrm{H}},(J \text { in } \mathrm{Hz}) \end{aligned}$ | $\delta_{\text {c }}$, type | $\begin{aligned} & 19 \\ & \delta_{\mathrm{H}},(J \text { in } \mathrm{Hz}) \end{aligned}$ | $\delta_{\text {c }}$, type | $\begin{aligned} & 20 \\ & \delta_{\mathrm{H}},(\mathrm{~J} \text { in } \mathrm{Hz}) \end{aligned}$ | $\delta_{\text {c }}$, type |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 4.72, d (3.9) | 100.0, CH | 4.78, d (3.8) | 100.0, CH | 5.04, d (3.8) | 96.8, CH | 5.06, d (3.9) | 96.9, CH |
| 2 | 3.48 , dd (10.2, 3.9) | 68.2, CH | 3.64 , dd (10.3, 3.8) | 68.4, CH | 4.98, dd (11.1, 3.8) | 68.2, CH | 5.00 , dd (11.2, 3.9) | 68.1, CH |
| 3 | 3.76 , dd (10.2, 3.6) | 67.1, CH | 3.78 , dd (10.3, 3.7) | 67.1, CH | 5.19, dd (11.1, 3.2) | 68.9, CH | 5.27 , dd (11.2, 3.2) | 68.8, CH |
| 4 | 5.11, d (3.6) | 74.0, CH | 5.16, d (3.7) | 74.0, CH | 4.01, d (3.2) | 76.5, CH | 4.03, d (3.2) | 76.3, CH |
| 5 | 4.08, q (6.6) | 64.4, CH | 4.11, q (6.6) | 64.5, CH | 4.06, q (6.6) | 65.9, CH | 4.08, q (6.6) | 66.1, CH |
| 6 | 0.96, d (6.6) | 16.0, $\mathrm{CH}_{3}$ | 0.97, d (6.6) | 16.0, $\mathrm{CH}_{3}$ | 1.14, d (6.6) | 16.0, $\mathrm{CH}_{3}$ | 1.15, d (6.6) | 16.2, $\mathrm{CH}_{3}$ |
| $2^{\prime}$ | 2.34, m | 62.0, CH |  | 62.1, CH | 2.09, m | 62.0, CH | 2.01, m | 61.8, CH |
| $3^{\prime}$ | 1.95, m | 30.0, $\mathrm{CH}_{2}$ | 1.99, m | 29.9, $\mathrm{CH}_{2}$ | 1.66, m | 29.7, $\mathrm{CH}_{2}$ | 1.66, m | 30.1, $\mathrm{CH}_{2}$ |
|  | 1.40, m |  | 1.44, m |  | 1.30, m |  | 1.28, m |  |
| $4^{\prime}$ | 1.61, m | 21.2, $\mathrm{CH}_{2}$ | 1.65, m | 21.2, $\mathrm{CH}_{2}$ | 1.57, m | 21.1, $\mathrm{CH}_{2}$ | 1.55, m | 21.4, $\mathrm{CH}_{2}$ |
|  |  |  |  |  | 1.45, m |  | 1.46, m |  |
| $5^{\prime}$ | 2.92, m | 56.2, $\mathrm{CH}_{2}$ | 2.98, m | 56.1, $\mathrm{CH}_{2}$ | 2.92, m | 56.2, $\mathrm{CH}_{2}$ | 2.86, m | 56.7, $\mathrm{CH}_{2}$ |
|  | 2.06, m |  | 2.16, m |  | 1.92, m |  | 1.87, m |  |
| $6^{\prime}$ | 1.84, m | 40.6, $\mathrm{CH}_{2}$ | 1.87, m | 40.3, $\mathrm{CH}_{2}$ | 1.87, m | 40.8, $\mathrm{CH}_{2}$ | 1.85, m | 41.5, $\mathrm{CH}_{2}$ |
|  | 1.20, m |  | 1.26, m |  | 1.13, m |  | 1.07, ddd (12.8, 10.0, 2.6) |  |
| $7{ }^{\prime}$ | 3.63, m | 74.3, CH | 3.65, m | 74.3, CH | 3.59, ddd (9.2, 6.2, 2.7) | 75.2, CH | 3.59, m | 75.5, CH |
| $8^{\prime}$ | 1.18, d (6.2) | 22.3, $\mathrm{CH}_{3}$ | 1.20, d (6.3) | 22.1, $\mathrm{CH}_{3}$ | 1.21, d (6.2) | 22.3, $\mathrm{CH}_{3}$ | 1.21, d (6.1) | 22.6, $\mathrm{CH}_{3}$ |
| $9^{\prime}$ | 2.19, s | 39.7, $\mathrm{CH}_{3}$ | $2.25, \mathrm{~s}$ | 39.6, $\mathrm{CH}_{3}$ | 2.15, s | 39.6, $\mathrm{CH}_{3}$ | 2.11, s | 39.8, $\mathrm{CH}_{3}$ |
| $1^{\prime \prime}$ |  | 165.4, C |  | 165.8, C |  | 166.5, C |  | 166.2, C |
| $2^{\prime \prime}$ | 6.06, d (12.8) | 119.4, CH | 6.66, d (16.0) | 118.1, CH |  | 127.6, C |  | 126.3, C |
| $3^{\prime \prime}$ | 7.01, d (12.8) | 141.9, CH | 7.65, d (16.0) | 144.2, CH | 6.73, m | 138.0, CH | 6.25 , qd (7.3, 1.6) | 140.4, CH |
| $4^{\prime \prime}$ |  | 134.5, C |  | 133.8, C | 1.73, m | 14.0, $\mathrm{CH}_{3}$ | 1.90 , dq (7.3, 1.6) | 15.8, $\mathrm{CH}_{3}$ |
| $5^{\prime \prime}$ | 7.66, dd (7.4, 2.2) | 129.6, CH | 7.74, m | 128.1, CH | 1.71, m | 11.6, $\mathrm{CH}_{3}$ | 1.77, t (1.6) | 20.2, $\mathrm{CH}_{3}$ |
| $6^{\prime \prime}$ | 7.35, m | 128.9, CH | 7.43, m | 130.3, CH |  |  |  |  |
| $7{ }^{\prime \prime}$ | 7.35, m | 127.8, CH | 7.43, m | 128.7, CH |  |  |  |  |
| $8^{\prime \prime}$ | 7.35, m | 128.9, CH | 7.43, m | 130.3, CH |  |  |  |  |
| $9^{\prime \prime}$ | 7.66, dd (7.4, 2.2) | 129.6, CH | 7.74, m | 128.1, CH |  |  |  |  |
| $1^{\prime \prime \prime}$ |  |  |  |  |  | 166.5, C |  | 166.5, C |
| $2^{\prime \prime \prime}$ |  |  |  |  |  | 127.6, C |  | 126.8, C |
| $3^{\prime \prime \prime}$ |  |  |  |  | 6.73, m | 138.0, CH | 6.17, qd (7.3, 1.6) | 139.2, CH |
| $4^{\prime \prime \prime}$ |  |  |  |  | 1.73, m | 14.0, $\mathrm{CH}_{3}$ | 1.85 , dq $(7.3,1.6)$ | 15.8, $\mathrm{CH}_{3}$ |
| $5^{\prime \prime \prime}$ |  |  |  |  | 1.71, m | 11.6, $\mathrm{CH}_{3}$ | 1.77, t (1.6) | 20.2, $\mathrm{CH}_{3}$ |
| $1^{\prime \prime \prime \prime}$ |  |  |  |  | 4.57, d (1.7) | 101.6, CH | 4.59, d (1.8) | 101.8, CH |
| $2^{\prime \prime \prime \prime}$ |  |  |  |  | 3.76, m | 70.1, CH | 3.76 , m | 70.3, CH |
| $3^{\prime \prime \prime \prime}$ |  |  |  |  | 3.48 , dd (9.4, 5.6, 3.2) | 70.1, CH | 3.47 , dd (9.4, 3.2) | 70.4, CH |
| $4^{\prime \prime \prime \prime}$ |  |  |  |  | 3.17 , td (9.4, 5.6) | 71.4, CH | 3.18 , t (9.4) | 71.7, CH |
| $5^{\prime \prime \prime \prime}$ |  |  |  |  | 3.65, m | 68.9, CH | 3.66, m | 69.1, CH |
| $6^{\prime \prime \prime \prime}$ |  |  |  |  | 0.91, d (6.1) | 17.5, $\mathrm{CH}_{3}$ | 0.95, d (6.2) | 18.0, $\mathrm{CH}_{3}$ |

Table 7
Antitrypanosomatid, antiplasmodial and cytotoxic activities of compounds 1-4, 6-18.

| Compound | T. b. rhodesiense | T. cruzi | L. donovani | P. falciparum | Cytotoxicity |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{IC}_{50}[\mu \mathrm{M}]$ |  |  |  |  |
| 1 | 49.3 | 88.0 | >100 | >100 | >100 |
| 2 | 80.9 | $>100$ | $>100$ | $>100$ | $>100$ |
| 3 | $>100$ | $>100$ | $>100$ | $>100$ | $>100$ |
| 4 | $>100$ | $>100$ | $>100$ | $>100$ | $>100$ |
| 6 | $>100$ | $>100$ | >100 | 94.0 | $>100$ |
| 7 | $>100$ | >100 | >100 | >100 | >100 |
| 8 | $>100$ | $>100$ | 66.2 | 27.2 | >100 |
| 9 | $>100$ | >100 | >100 | 65.7 | >100 |
| 10 | >100 | $>100$ | 68.9 | >100 | $>100$ |
| 11 | 25.7 | $>100$ | 94.6 | 6.8 | $>100$ |
| 12 | 23.7 | $>100$ | 81.8 | 7.3 | $>100$ |
| 13 | 21.2 | $>100$ | >100 | 6.3 | >100 |
| 14 | 17.5 | $>100$ | >100 | 8.7 | $>100$ |
| 15 | 11.6 | $>100$ | 53.2 | 2.8 | $>100$ |
| 16 | 13.1 | >100 | 91.1 | 6.0 | >100 |
| 17 | 94.5 | $>100$ | >100 | 26.1 | $>100$ |
| 18 | 18.9 | >100 | >100 | 6.3 | $>100$ |
| Schizanthus extract ${ }^{\text {a }}$ | 31.8 | 65.0 | 59.6 | 8.9 | >100 |
| Melarsoprol ${ }^{\text {b }}$ | 0.013 |  |  |  |  |
| Benznidazole ${ }^{\text {b }}$ |  | 2.8 |  |  |  |
| Miltefosine ${ }^{\text {b }}$ |  |  | 0.55 |  |  |
| Chloroquine ${ }^{\text {b }}$ |  |  |  | 0.006 |  |
| Podophyllotoxin ${ }^{\text {b }}$ |  |  |  |  | 0.027 |

[^1]Hinds Instruments, Hillsboro, OR, USA) set at l/4 retardation was used to modulate the handedness of the circular polarized light. Demodulation was performed by a lock-in amplifier (SR830 DSP, Stanford Research System, Sunnyvale, CA, USA). An optical low-pass filter $\left(<1800 \mathrm{~cm}^{-1}\right.$ ) in front of the photoelastic modulator was used to enhance the signal/noise ratio. Solutions of $2.5-4 \mathrm{mg}$ in $170 \mu \mathrm{~L}$ of DMSO- $d_{6}$ were prepared and measured in a transmission cell equipped with $\mathrm{CaF}_{2}$ windows and a $200 \mu \mathrm{~m}$ spacer. Artifacts were eliminated by subtracting the VCD spectrum of the pure solvent (reference) from the VCD spectrum of the compound. For both the sample and the reference, ca. 24,000 scans at $4 \mathrm{~cm}^{-1}$ resolution were averaged. UV spectra were recorded on a PerkinElmer Lambda-25 UV-vis spectrophotometer (Wellesley, MA, USA). IR spectra were measured on a PerkinElmer Spectrum 100 spectrometer. NMR spectra were recorded on a Bruker Avance III HD 600 MHz NMR spectrometer equipped with a QCI 5 mm Cryoprobe and a SampleJet automated sample changer (Bruker BioSpin, Rheinstetten, Germany). Chemical shifts are reported in parts per million ( $\delta$ ) using the residual $\mathrm{CD}_{3} \mathrm{OD}$ signals ( $\delta_{\mathrm{H}} 3.31$; $\delta_{\mathrm{C}} 49.0$ ) or DMSO- $d_{6}$ signal ( $\delta_{\mathrm{H}} 2.50 ; \delta_{\mathrm{C}} 39.5$ ) as internal standards for ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, respectively. HRMS spectra were obtained on a Q Exactive Focus Hybrid quadripole-orbitrap mass spectrometer (Thermo Scientific, Waltham, MA, USA) using electrospray in the positive mode. The spray voltage was set to 3.5 kV ; the sheath gas flow rate $\left(\mathrm{N}_{2}\right)$ to 50 units; the capillary temperature to $320^{\circ} \mathrm{C}$; the S lens RF level to 50 ; and the probe heater temperature to $425{ }^{\circ} \mathrm{C}$. UHPLC was performed on an Acquity UPLC I-class System (Waters, Milford, MA, USA). The separation was performed on an Acquity BEH C $\mathrm{C}_{18}$ UPLC column ( $150 \times 2.1 \mathrm{~mm}$ i.d.; 1.7 $\mu \mathrm{m}$, Waters), using a gradient $\left(\mathrm{H}_{2} \mathrm{O}\right.$ and MeCN both containing $0.1 \%$ formic acid) of $5-30 \% \mathrm{MeCN}$ in 35 min , followed by a washing step with $98 \% \mathrm{MeCN}$ for 2 min . After the washing step, the column was equilibrated with $5 \% \mathrm{MeCN}$ for 5 min before the next injection. The flow rate was set to $0.4 \mathrm{~mL} / \mathrm{min}$, the temperature to $40^{\circ} \mathrm{C}$, and the injection volume was $1 \mu \mathrm{~L}$. Flash chromatography was performed on an Armen Spot preparative chromatographic system (Interchim, Montluçon, France) equipped with a quaternary pump, a UV detector, and a fraction collector. Reverse phase semi-preparative chromatography was performed on an Armen Spot preparative chromatographic system with a Kinetex Axia Core Shell $\mathrm{C}_{18}$ column ( $250 \times 21.2 \mathrm{~mm} ; 5 \mu \mathrm{~m}$, Phenomenex) or an X-Select CSH C 18 column ( $250 \times 19 \mathrm{~mm}, 5 \mu \mathrm{~m}$; Waters). The flow rate was set to 20 and $18 \mathrm{~mL} / \mathrm{min}$, respectively, and UV absorbance was at 220 nm . The control of fractions was carried out by an Acquity UPLC System (Waters) equipped with an Acquity PDA detector and connected to a Quattro Micro triple quadrupole mass spectrometer (Waters) with an ESI source operating in positive mode.

### 4.2. NMR computational details

All calculations have been performed with the Gaussian 16 program (Frisch et al., 2016). Geometry optimization has been achieved using density function theory (DFT) with the B3LYP functional and the 6-31G (d) basis set in the gas-phase. Vibrational analysis was completed at the same level to confirm a minimum. NMR prediction was performed using the mPW1PW91/6-31 $+\mathrm{g}(\mathrm{d}, \mathrm{p})$ level.

### 4.3. Plant material

The aerial parts of Schizanthus tricolor Grau \& Gronbach (Solanaceae) were collected in December 2003 in Cachagua, Chile (GPS coordinates $32^{\circ} 59^{\prime} \mathrm{S} ; 71^{\circ} 44^{\prime} \mathrm{W}$ ). Their identification was confirmed by Prof. Fernanda Pérez (Departamento de Botanica, Universidad de Chile). A voucher specimen was deposited at the Facultad de Ciencas Quimicas ( $\mathrm{N}^{\circ}$ 2000-3).

### 4.4. Extraction and isolation

The plant material ( 1.3 kg ) was extracted successively with hexane
and MeOH at room temperature. After filtration, the alcoholic solution was evaporated to dryness. The residue ( 161.2 g ) was taken up in 0.1 M HCl and extracted with $\mathrm{Et}_{2} \mathrm{O}$. The aqueous solution was basified with $4 \%$ $\mathrm{NH}_{4} \mathrm{OH}$ to pH 12 and then extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic solvent was dried with anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and evaporated, yielding 3.7 g of a gummy alkaline residue. The residue was submitted to a fractionation using two flash columns connected in series ( $\mathrm{PF}-\mathrm{C}_{18} \mathrm{HQ}$ / $120 \mathrm{~g}, 15 \mu \mathrm{~m} \mathrm{C}_{18}$, Interchim) with a gradient of $5-40 \% \mathrm{MeCN}+0.1 \%$ FA in 2 h and afforded 35 fractions. Fractions 1 and 2 were grouped ( 91.5 mg ) and fractionated with the Kinetex column using a gradient of $0-15 \%$ $\mathrm{MeOH}+0.1 \% \mathrm{NH}_{4} \mathrm{OH}$ in 90 min , and yielded $1(15.1 \mathrm{mg}), 2(6.0 \mathrm{mg})$, and $3(2.5 \mathrm{mg})$. Fraction $3(23.9 \mathrm{mg})$ was separated with the Kinetex column using a gradient of $5-20 \% \mathrm{MeOH}+0.1 \% \mathrm{NH}_{4} \mathrm{OH}$ in 60 min and afforded 4 ( 2.3 mg ) and $21(0.3 \mathrm{mg})$. Fraction $4(40.8 \mathrm{mg})$ was separated with the X-Select column using a gradient of $5-25 \% \mathrm{MeOH}+0.1 \%$ $\mathrm{NH}_{4} \mathrm{OH}$ in 60 min and gave $5(0.7 \mathrm{mg}), 6(1.4 \mathrm{mg}), 7(1.0 \mathrm{mg})$ and 22 ( 0.3 mg ). Fraction 5 ( 28.4 mg ) was separated with the Kinetex column using a gradient of $10-30 \% \mathrm{MeOH}+0.1 \% \mathrm{NH}_{4} \mathrm{OH}$ in 60 min and afforded 8 ( 3.0 mg ), $23(0.2 \mathrm{mg})$ and $24(0.3 \mathrm{mg})$. Fraction $6(40.3 \mathrm{mg})$ was separated with the Kinetex column using a gradient of 10-30\% ACN in 60 min and afforded $9(2.6 \mathrm{mg}), 25(0.2 \mathrm{mg})$ and $26(0.2 \mathrm{mg})$. Fraction 7 ( 35.4 mg ) was separated with the Kinetex column using a gradient of $15-35 \% \mathrm{MeOH}+0.1 \% \mathrm{NH}_{4} \mathrm{OH}$ in 60 min and gave $10(1.0 \mathrm{mg})$ and 17 ( 0.9 mg ). Fraction $8(30.4 \mathrm{mg})$ was separated with the Kinetex column using a gradient of $15-40 \% \mathrm{MeOH}+0.1 \% \mathrm{NH}_{4} \mathrm{OH}$ in 60 min and gave 19 ( 0.5 mg ). Fraction $10(20.8 \mathrm{mg})$ was separated with the X-Select column using a gradient of $15-40 \% \mathrm{MeOH}+0.1 \% \mathrm{NH}_{4} \mathrm{OH}$ in 60 min and gave $20(0.9 \mathrm{mg})$. Fraction $11(25.9 \mathrm{mg})$ was separated with the Kinetex column using a gradient of $15-45 \% \mathrm{MeOH}+0.1 \% \mathrm{NH}_{4} \mathrm{OH}$ in 60 min and gave 18 ( 1.3 mg ). Fractions 12 and 13 were grouped ( 45.8 mg ) and fractionated with the Kinetex column using a gradient of $15-40 \%$ ACN in 60 min and yielded 11 ( 3.8 mg ). Fraction 15 ( 16.6 mg ) was separated with the Kinetex column using a gradient of $20-45 \%$ $\mathrm{MeOH}+0.1 \% \mathrm{NH}_{4} \mathrm{OH}$ in 60 min and afforded $12(4.0 \mathrm{mg})$. Fraction 18 ( 40.6 mg ) was separated with the Kinetex column using a gradient of $20-45 \% \mathrm{MeOH}+0.1 \% \mathrm{NH}_{4} \mathrm{OH}$ in 60 min and gave $13(1.6 \mathrm{mg})$. Fraction 22 ( 32.0 mg ) was separated with the X-Select column using a gradient of $25-50 \% \mathrm{MeOH}+0.1 \% \mathrm{NH}_{4} \mathrm{OH}$ in 60 min and gave 14 (1.1 $\mathrm{mg})$. Fraction $25(28.0 \mathrm{mg})$ was separated with the X-Select column using a gradient of $30-55 \% \mathrm{MeOH}+0.1 \% \mathrm{NH}_{4} \mathrm{OH}$ in 60 min and gave 15 ( 3.8 mg ). Fraction $30(25.0 \mathrm{mg}$ ) was separated with the X-Select column using a gradient of $30-55 \% \mathrm{MeOH}+0.1 \% \mathrm{NH}_{4} \mathrm{OH}$ in 60 min and gave $16(2.5 \mathrm{mg})$.

### 4.4.1. Schizanthoside A1 (1)

Pale yellow oil; $[a]_{D}^{22}-33.2(c 0.05, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max } 222 \mathrm{~nm}$; IR (neat) $\nu_{\max } 3255,2907,1592,1458,1377,1342 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Table 1; HRESIMS $m / z 290.1979[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{14} \mathrm{H}_{27} \mathrm{NO}_{5}, 290.1962$ ).

### 4.4.2. Schizanthoside A2 (2)

Pale yellow oil; $[a]_{D}^{22}-56.7$ (c 0.05, MeOH); UV (MeOH) $\lambda_{\max } 222 \mathrm{~nm}$; IR (neat) $\nu_{\max } 3305,2923,1596,1452,1345,1050 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Table 1; HRESIMS $m / z 290.1979[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{14} \mathrm{H}_{27} \mathrm{NO}_{5}, 290.1962$ ).

### 4.4.3. Schizanthoside A3 (3)

Pale yellow oil; $[a]_{D}^{22}-40.3$ (c 0.07, MeOH); UV (MeOH) $\lambda_{\text {max }} 222 \mathrm{~nm}$; IR (neat) $\nu_{\max } 3293,2912,1593,1455,1375,1345 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Table 1; HRESIMS $m / z 290.1979[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{14} \mathrm{H}_{27} \mathrm{NO}_{5}, 290.1962$ ).

### 4.4.4. Schizanthoside B1 (4)

Pale yellow oil; $[a]_{D}^{22}-78.6$ (c 0.07, MeOH); UV (MeOH) $\lambda_{\max } 220 \mathrm{~nm}$; IR (neat) $\nu_{\max } 3325,2934,1711,1599,1457,1376 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$

NMR, see Tables 2 and 3; HRESIMS $m / z 372.2385[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{19} \mathrm{H}_{34} \mathrm{NO}_{6}, 372.2381$ ).

### 4.4.5. Schizanthoside B2 (5)

Pale yellow oil; $[a]_{D}^{22}$ and IR not measured due to lack of material; UV $(\mathrm{MeOH}) \lambda_{\max } 220 \mathrm{~nm} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Tables 2 and 3; HRESIMS m/ z $372.2385[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{19} \mathrm{H}_{34} \mathrm{NO}_{6}, 372.2381$ ).

### 4.4.6. Schizanthoside B3 (6)

Pale yellow oil; $[a]_{D}^{22}-71.6$ (c 0.05, MeOH); UV (MeOH) $\lambda_{\max } 220 \mathrm{~nm} ;$ IR (neat) $\nu_{\max } 3351,2929,1706,1595,1457,1353 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Tables 2 and 3; HRESIMS $m / z 372.2385[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{19} \mathrm{H}_{34} \mathrm{NO}_{6}, 372.2381$ ).

### 4.4.7. Schizanthoside B4 (7)

Pale yellow oil; $[a]_{D}^{22}-81.0(c 0.05, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max } 220 \mathrm{~nm} ;$ IR (neat) $\nu_{\max } 3352,2970,2927,1714,1650,1599,1449,1142,1082$, $1043 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Tables 2 and 3; HRESIMS $\mathrm{m} / \mathrm{z}$ $372.2385[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{19} \mathrm{H}_{34} \mathrm{NO}_{6}, 372.2381$ ).

### 4.4.8. Schizanthoside B5 (8)

Pale yellow oil; $[a]_{D}^{22}-62.7$ (c 0.05, MeOH); UV (MeOH) $\lambda_{\max } 220 \mathrm{~nm} ;$ IR (neat) $\nu_{\max } 3360,2936,1713,1597,1457,1352 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Tables 2 and 3; HRESIMS $m / z 372.2385[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{19} \mathrm{H}_{34} \mathrm{NO}_{6}, 372.2381$ ).

### 4.4.9. Schizanthoside B6 (9)

Pale yellow oil; $[a]_{D}^{22}-72.2$ (c 0.05, MeOH); UV (MeOH) $\lambda_{\max } 220 \mathrm{~nm}$; IR (neat) $\nu_{\max } 3344,2927,1712,1597,1453,1380 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Tables 2 and 3; HRESIMS $m / z 372.2385[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{19} \mathrm{H}_{34} \mathrm{NO}_{6}, 372.2381$ ).

### 4.4.10. Schizanthoside B7 (10)

Pale yellow oil; $[a]_{D}^{22}-91.8$ (c 0.05, MeOH); UV (MeOH) $\lambda_{\max } 220 \mathrm{~nm} ;$ IR (neat) $\nu_{\max } 3307,2935,1706,1650,1597,1449,1026 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and
${ }^{13} \mathrm{C}$ NMR, see Tables 2 and 3; HRESIMS $m / z 372.2385[\mathrm{M}+\mathrm{H}]{ }^{+}$(calcd for $\mathrm{C}_{19} \mathrm{H}_{34} \mathrm{NO}_{6}, 372.2381$ ).

### 4.4.11. Schizanthoside C1 (11)

Pale yellow oil; $[a]_{D}^{22}-65.1$ (c 0.07, MeOH); UV (MeOH) $\lambda_{\text {max }} 221 \mathrm{~nm}$; IR (neat) $\nu_{\max } 3376,2934,1713,1600,1457,1354 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Tables 4 and 5; HRESIMS $m / z 454.2797[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{24} \mathrm{H}_{40} \mathrm{NO}_{7}, 454.2800$ ).

### 4.4.12. Schizanthoside C2 (12)

Pale yellow oil; $[a]_{D}^{22}-65.4$ (c 0.07, MeOH); UV (MeOH) $\lambda_{\max } 221 \mathrm{~nm} ;$ IR (neat) $\nu_{\max } 3341,2934,1713,1602,1457,1356 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Tables 4 and 5; HRESIMS $m / z 454.2797[M+H]^{+}$(calcd for $\mathrm{C}_{24} \mathrm{H}_{40} \mathrm{NO}_{7}, 454.2800$ ).

### 4.4.13. Schizanthoside C3 (13)

Pale yellow oil; $[a]_{D}^{22}-32.2(c 0.05, \mathrm{MeOH})$; UV (MeOH) $\lambda_{\max } 221 \mathrm{~nm} ;$ IR (neat) $\nu_{\max } 3351,2970,1723,1602,1449,1367 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Tables 4 and 5; HRESIMS $m / z 454.2797[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{24} \mathrm{H}_{40} \mathrm{NO}_{7}, 454.2800$ ).

### 4.4.14. Schizanthoside C4 (14)

Pale yellow oil; $[a]_{D}^{22}-46.7$ (c 0.05, MeOH); UV (MeOH) $\lambda_{\text {max }} 221 \mathrm{~nm}$; IR (neat) $\nu_{\max } 3351,2927,1713,1596,1453,1037 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Tables 4 and 5; HRESIMS $m / z 454.2797[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{24} \mathrm{H}_{40} \mathrm{NO}_{7}, 454.2800$ ).

### 4.4.15. Schizanthoside C5 (15)

Pale yellow oil; $[a]_{D}^{22}-46.5$ (c 0.07, MeOH); UV (MeOH) $\lambda_{\max } 221 \mathrm{~nm} ;$

IR (neat) $\nu_{\max } 3353,2935,1719,1595,1456,1377 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Tables 4 and 5; HRESIMS $m / z 454.2797[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{24} \mathrm{H}_{40} \mathrm{NO}_{7}, 454.2800$ ).

### 4.4.16. Schizanthoside C6 (16)

Pale yellow oil; $[a]_{D}^{22}-45.7$ (c 0.07, MeOH); UV (MeOH) $\lambda_{\max } 221 \mathrm{~nm}$; IR (neat) $\nu_{\max } 3353,2970,1739,1456,1368 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Tables 4 and 5; HRESIMS $m / z 454.2797[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{24} \mathrm{H}_{40} \mathrm{NO}_{7}$, 454.2800).

### 4.4.17. Schizanthoside D1 (17)

Pale yellow oil; $[a]_{D}^{22}-43.2$ (c 0.05, MeOH); UV (MeOH) $\lambda_{\max } 221$, 282 nm ; IR (neat) $\nu_{\max } 3327,2940,2831,1709,1601,1450,1023 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Table 6; HRESIMS $m / z 420.2388[\mathrm{M}+\mathrm{H}]{ }^{+}$(calcd for $\mathrm{C}_{23} \mathrm{H}_{34} \mathrm{NO}_{6}, 420.2381$ ).
4.4.18. Schizanthoside D2 (18)

Pale yellow oil; $[a]_{D}^{22}-74.3$ (c 0.05, MeOH); UV (MeOH) $\lambda_{\max } 220$, 282 nm ; IR (neat) $\nu_{\max } 3350,2939,2830,1715,1597,1450,1024 \mathrm{~cm}^{-1}$;
${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Table 6; HRESIMS $m / z 420.2388[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{23} \mathrm{H}_{34} \mathrm{NO}_{6}, 420.2381$ ).

### 4.4.19. Schizanthoside E1 (19)

Pale yellow oil; $[a]_{D}^{22}$ not measured due to lack of material; UV (MeOH) $\lambda_{\max } 221 \mathrm{~nm}$; IR (neat) $\nu_{\max } 3349,2930,1711,1597,1454$, $1024 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Table 6; HRESIMS $\mathrm{m} / \mathrm{z} 600.3386[\mathrm{M}+$ $\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{30} \mathrm{H}_{50} \mathrm{NO}_{11}, 600.3379$ ).

### 4.4.20. Schizanthoside E2 (20)

Pale yellow oil; $[a]_{D}^{22}$ not measured due to lack of material; UV $(\mathrm{MeOH}) \lambda_{\max } 221 \mathrm{~nm}$; IR (neat) $\nu_{\text {max }} 3351,2937,1708,1596,1452$, $1024 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Table 6; HRESIMS $m / z 600.3386[\mathrm{M}+$ $\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{30} \mathrm{H}_{50} \mathrm{NO}_{11}, 600.3379$ ).

### 4.5. Acid hydrolysis of 1 and 2

A solution of 1 and 2 ( 3.0 mg each, individually) in $10 \% \mathrm{HCl}(2 \mathrm{~mL})$ was heated for 3 h . Then, the solution was basified with NaOH ( 2 M ) until pH 12 , and then extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 4 \mathrm{~mL})$. (-)-Hygroline and $(+)$-pseudohygroline were confirmed by comparison of their optical rotation with the literature (Liniger et al., 2013) or by comparison with a reference sample for the L-fucose; $[\alpha]_{D}^{20}-75.5$ (c 4, $\mathrm{H}_{2} \mathrm{O}$ ).
4.6. Antitrypanosomal, antileishmanial, antiplasmodial, and cytotoxicity assays

The in vitro activity against T. b. rhodesiense, T. cruzi, L. donovani and P. falciparum as well as cytotoxicity assessment in L6 cells were determined as reported elsewhere (Bernal et al., 2020). The strains were T. b. rhodesiense STIB 900 bloodstream form (trypomastigote), T. cruzi Tulahuen C2C4 (LacZ) amastigote form grown in rat L6 skeletal myoblasts, L. donovani MHOM/ET/67/L82 amastigote form, and P. falciparum NF54 erythrocytic stage. Results are expressed in $\mu \mathrm{M}$ for pure compounds.

## Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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## Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi. org/10.1016/j.phytochem.2021.112957.

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[^1]:    Results are the means of two independent assays.
    ${ }^{\text {a }}{ } \mathrm{IC}_{50}$ is expressed in $\mu \mathrm{g} / \mathrm{mL}$.
    ${ }^{\mathrm{b}}$ Positive controls.

