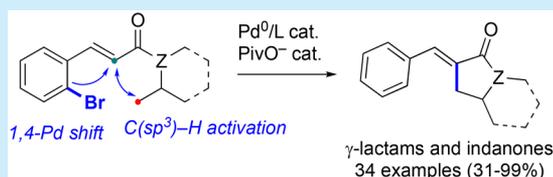


1,4-Palladium Shift/C(sp³)–H Activation Strategy for the Remote Construction of Five-Membered Rings

Ronan Rocaboy and Olivier Baudoin*[Ⓜ]

University of Basel, Department of Chemistry, St. Johanns-Ring 19, CH-4056 Basel, Switzerland

ABSTRACT: 1,*n*-Metal shift is an elegant alternative approach enabling the functionalization of remote C–H bonds from simple precursors. In this work, we report a novel and simple Pd⁰-catalyzed domino reaction involving 1,4-palladium shift and C(sp³)–H activation and leading to (fused) five-membered rings. This method allowed access to a broad range of valuable arylidene γ -lactams and indanones and was applied to the formal synthesis of (–)-pyrrolam.

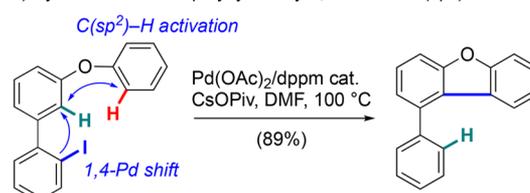


The last two decades witnessed impressive developments in the formation of carbon–carbon and carbon–heteroatom bonds by transition-metal-catalyzed C–H activation, generally affording improved atom- and step-economy compared to traditional cross-coupling methods.¹ In addition to direct C–H functionalization methods, strategies based on 1,*n*-metal shift allow the functionalization of distal C–H bonds which may be otherwise difficult to access.² Since the initial observation of 1,4-palladium shift by Heck in 1972,³ a number of 1,*n*-Pd migrations occurring between a wide range of C(sp²)- or C(sp³)-hybridized carbon atoms have been reported.² In 2003, Larock and co-workers showed the first example of Pd⁰-catalyzed domino reaction⁴ involving oxidative addition, 1,4-Pd shift, and C(sp²)–H arylation, resulting in the construction of complex polycyclic molecules (Scheme 1a).⁵ Later, they reported a domino reaction involving 1,4-Pd shift, carbopalladation, and C(sp³)–H activation to form a fused cyclopropane.⁶ A few years later, Zhu and co-workers described a general method to access fused oxindoles by combining carbopalladation, 1,4-Pd shift, and activation of benzylic C(sp³)–H bonds (Scheme 1b).⁷ However, to the best of our knowledge, there is no example of a method simply combining oxidative addition, 1,4-Pd shift, and C(sp³)–H activation without an intermediate carbopalladation step. In the past years, our group has developed a set of Pd⁰-catalyzed methods for the direct functionalization of C(sp³)–H bonds from precursors containing a C(sp²)–X bond (X = leaving group).⁸ In particular, we reported the synthesis of (fused) γ -lactams from alkenyl bromides (Scheme 1c).⁹ To extend the scope of this reaction, we hypothesized that the organopalladium intermediate arising from C–Br oxidative addition might be also generated by 1,4-Pd shift from a more remote C–X bond. Such an indirect strategy would allow the use of less congested, easily accessible substrates.

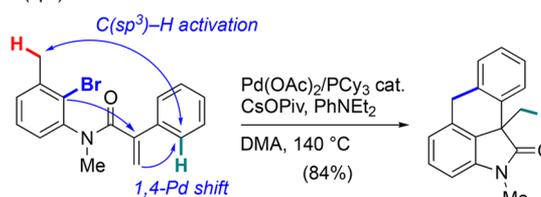
A mechanistic blueprint for this domino process is depicted in Scheme 2. Oxidative addition from aryl bromide **1** followed by bromide–carboxylate exchange leads to organopalladium intermediate **A**. Subsequent C(sp²)–H activation through the carboxylate-mediated concerted metalation–deprotonation

Scheme 1. 1,4-Pd Shift/C–H Activation and Synthesis of γ -Lactams

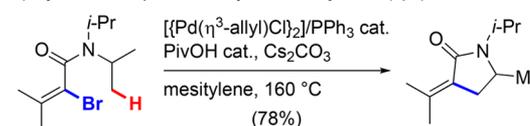
a) Synthesis of fused polycycles by 1,4-Pd shift/C(sp²)–H activation



b) Synthesis of oxindoles by carbopalladation/1,4-Pd shift/C(sp³)–H activation

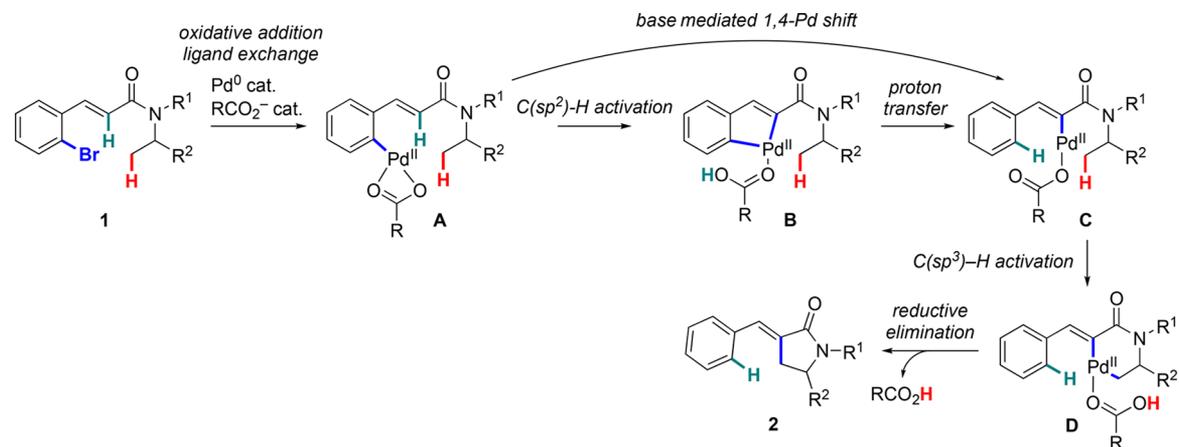


c) Synthesis of γ -lactams by Pd⁰-catalyzed C(sp³)–H activation



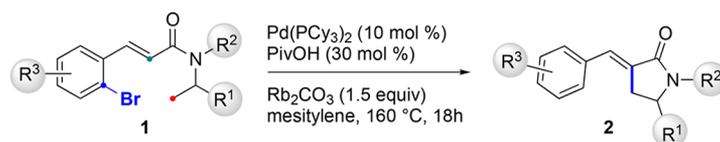
mechanism¹⁰ affords the 5-membered palladacycle **B**. The latter is too strained to undergo reductive elimination and should readily open by proton transfer from the coordinated carboxylic acid, according to previous experimental observations¹¹ and mechanistic studies,¹² to give intermediate **C**. These two steps from **A** to **C** result in the net aryl to vinyl 1,4-Pd shift.^{13,14} Organopalladium **C** is the same intermediate formed during the previous direct C(sp³)–H activation reaction (see Scheme 1c).⁹ Hence, base-mediated C(sp³)–H activation from

Scheme 2. Mechanistic Hypothesis^a

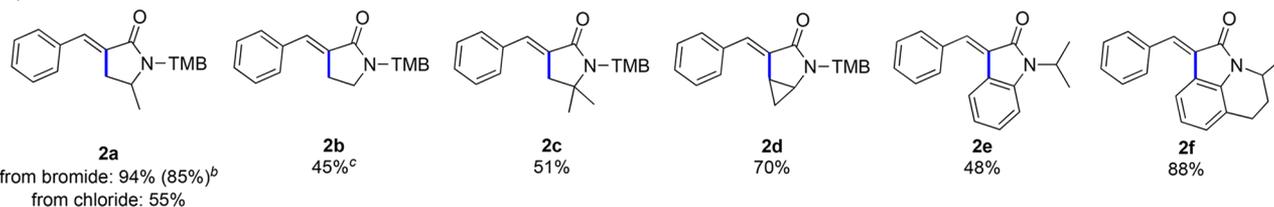


^aThe ligand has been omitted for clarity.

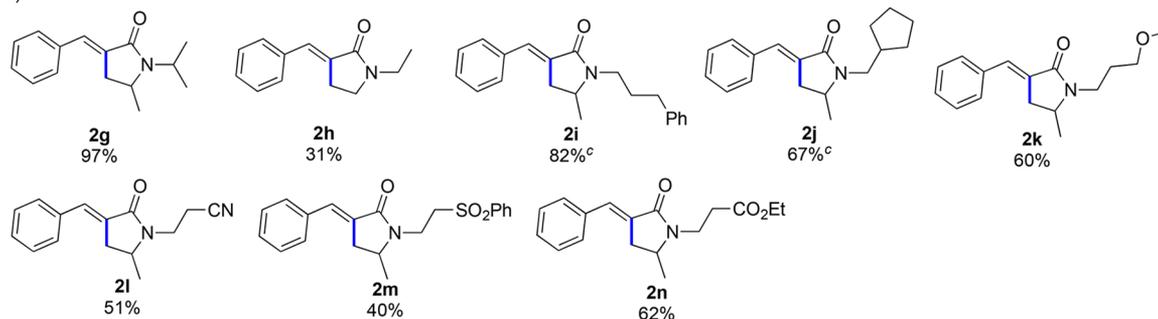
Scheme 3. Scope of the Synthesis of α -Arylidene γ -Lactams^a



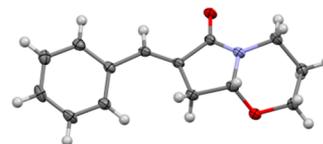
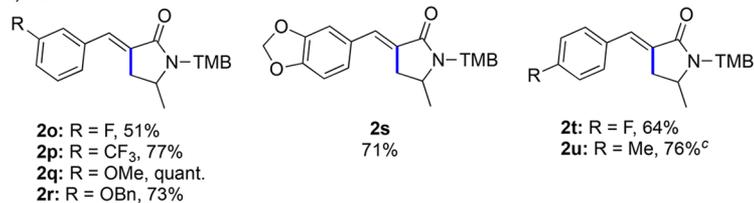
a) Variation of R¹



b) Variation of R²

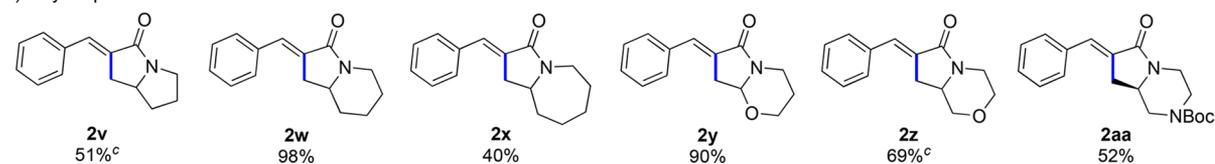


c) Variation of R³



X-ray structure of **2y^d**

d) Bicyclic products



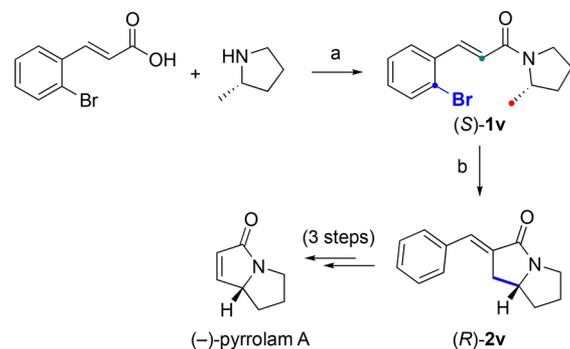
^aAll reactions were performed on a 0.1 mmol scale unless otherwise noted. ^bPerformed on a 1 mmol scale. ^cUsing additional PCy₃ (10 mol %). ^dThermal ellipsoids shown at 50% probability. TMB = 2,4,6-trimethoxybenzyl.

C and reductive elimination from the resulting 6-membered palladacycle **D** would lead to γ -lactam **2**. At this point, we were aware of potential pitfalls resulting from the lack of precedence for (1) 1,4-Pd shift onto the α,β -unsaturated system and (2) the combination of 1,4-Pd shift with the activation of nonactivated C(sp³)-H bonds. Herein, we report the development of such a domino reaction to access a wide range of arylidene γ -lactams and indanones.

We started our investigations with the synthesis of γ -lactams **2** (Scheme 3). The optimization of reaction conditions was performed on the TMB-protected¹⁵ isopropylamide **1a** derived from 2-bromocinnamic acid (Table S1). The desired product **2a** was obtained in 94% yield on a 0.1 mmol scale using the well-defined complex Pd(PCy₃)₂ as the catalyst,¹⁶ co-catalytic pivalic acid, and Rb₂CO₃ as the stoichiometric base in mesitylene at 160 °C. This high temperature was required, similar to our previous study on the direct reaction,⁹ to favor the formation of the strained α -arylidene γ -lactam and avoid the protodebromination side reaction. The reaction also proceeded satisfyingly on a 10-fold (1 mmol) scale, giving rise to **2a** in 85% yield. With the optimized conditions in hand, we studied the scope of the reaction. Addition of free PCy₃ (10 mol %) was found to be beneficial in some cases, presumably to avoid catalyst decomposition. Using the aryl chloride instead of the bromide also furnished **2a**, albeit in lower yield (55%). The influence of the alkyl group undergoing C-H activation was first studied on amides containing the TMB group (Scheme 3a). Average to good yields were achieved for ethyl (**2b**), *tert*-butyl (**2c**), as well as cyclopropyl (**2d**) groups. The former is a challenging case due to the lesser number of methyl groups and the lack of a Thorpe-Ingold effect favoring the C(sp³)-H activation step. Expectedly, the competition between C(sp²)-H and C(sp³)-H activation was clearly in favor of the former, giving rise to the interesting (fused) oxindoles **2e,f**. Substrates bearing two potentially reactive substituents on the amide nitrogen were next examined (Scheme 3b). Average to very good yields were observed, together with a high site-selectivity for the primary positions of the isopropyl group vs equidistant secondary positions (**2i**, **2k-n**), including much more acidic ones adjacent to nitrile, sulfone, and ester groups (**2l-n**). It should be noted that the β -lactam arising from C-H activation at the α -position to the nitrogen atom^{9,17} was never observed (e.g., **2b**, **2d**, **2h**). Next, the effect of substituents on the aromatic ring was studied (Scheme 3c). Electron-withdrawing or -donating groups at the *meta*- or *para*-position to the bromine atom were well tolerated, furnishing the corresponding products with good to excellent yields (**2o-u**). Interestingly, such α -arylidene γ -lactams have been shown to exhibit antifungal activities toward *Colletotrichum orbiculare*.¹⁸ Finally, we turned our attention to the synthesis of bicyclic γ -lactams (Scheme 3d). The fused pyrrolidine **2v** and azepane **2x**, relevant to the synthesis of pyrrolizidine^{19a} and *Stemona* alkaloids,^{19b} respectively, were obtained from easily available precursors in 40–51% yield. In contrast, the fused piperidine **2w** was obtained in much higher yield (98%). This result was successfully extended to bicyclic 5,6-fused γ -lactams containing heteroatoms, such as oxazinanes **2y-z** and the enantiopure *N*-Boc-protected piperazine **2aa**. Of note, olefin isomerization of the reaction products was never observed.

Its application to the short formal synthesis of (-)-pyrrolam A, a pyrrolizidine alkaloid isolated from *Streptomyces olivaceus* strains,²⁰ illustrates the simplicity of the current method

Scheme 4. Formal Synthesis of (-)-Pyrrolam A^a

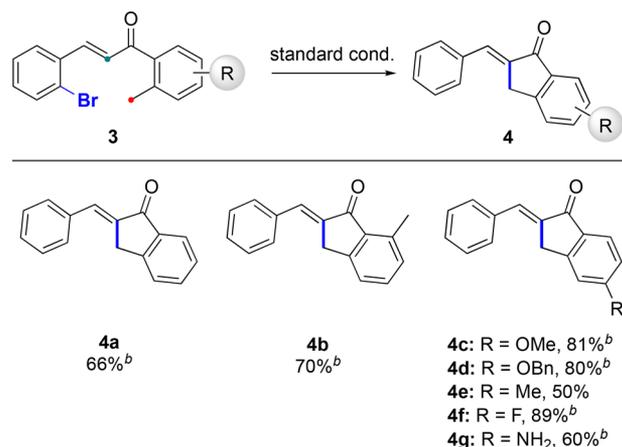


^aReaction conditions: (a) (COCl)₂ (1.5 equiv), Et₃N (2 equiv), CH₂Cl₂, 20 °C, quant; (b) Pd(PCy₃)₂ (10 mol %), PCy₃ (10 mol %), PivOH (30 mol %), Rb₂CO₃ (1.5 equiv), mesitylene (*c* = 0.025M), 160 °C, 18 h, 50%.

(Scheme 4). Standard amide formation from (*S*)-2-methylpyrrolidine and 2-bromocinnamic acid, both commercially available, gave the precursor for the key 1,4-Pd shift/C(sp³)-H activation reaction (**1v**). The latter was reacted under standard conditions to afford the enantiopure γ -lactam **2v** in 50% yield. Compound **2v** was previously converted to (-)-pyrrolam A in three steps;²¹ hence, the current approach allows for the synthesis of pyrrolam A in only five steps.

Next, we turned our attention toward the extension of the current method to other α,β -unsaturated carbonyl substrates for which the direct C(sp³)-H activation reaction is not known. In particular, we examined the reactivity of readily available chalcones **3** containing benzylic C(sp³)-H bonds (Scheme 5).²² The reaction proceeded remarkably well under

Scheme 5. Synthesis of Arylidene Indanones^a



^aReaction conditions: see Scheme 3. ^bUsing additional PCy₃ (10 mol %).

the standard conditions, thereby furnishing a range of arylidene indanones **4**. The reaction was compatible with electron-rich and electron-deficient substituents (**4b-f**), and even with a free aniline (**4g**), giving rise to the corresponding products in average to excellent yields (50–89%).²³ Of note, such compounds possess a variety of interesting biological properties.²⁴

To gain mechanistic insights, we performed experiments with fully and partially deuterated substrate **1a**, bearing deuterium atoms on the key C(sp²) and C(sp³) positions undergoing C-H activation (Scheme S1). We observed an

unexpectedly strong intermolecular D–H exchange,^{12b} preventing us from analyzing the 1,4-Pd shift, but indicating that the C(sp³)–H activation step (Scheme 2, C → D) is reversible and faster than the final reductive elimination leading to the strained α -arylidene γ -lactam ring.^{9,16c}

In conclusion, we reported a simple, step-economical method to construct (fused) five-membered rings through a novel Pd⁰-catalyzed domino reaction involving 1,4-palladium shift and C(sp³)–H activation. The generality of this method was demonstrated on a broad range of arylidene γ -lactams and indanones, and its applicability was illustrated through the formal synthesis of (–)-pyrrolam. This work opens the way to the development of C(sp³)–H functionalization reactions that are difficult to achieve through direct methods.

■ ASSOCIATED CONTENT

Accession Codes

CCDC 1884100 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: olivier.baudoin@unibas.ch.

ORCID

Olivier Baudoin: 0000-0002-0847-8493

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

This work was financially supported by the University of Basel. We thank Dr. D. Häussinger, University of Basel, for NMR experiments, Dr. Alessandro Prescimone, University of Basel, for X-ray diffraction analysis and Dr. M. Pfeffer, University of Basel, for MS analyses.

■ REFERENCES

- (1) (a) *Catalytic Transformations via C–H Activation*; Yu, J.-Q., Ed.; Science of Synthesis; Georg Thieme Verlag KG, Stuttgart, 2015; Vols. 1 and 2. (b) Hartwig, J. F. *J. Am. Chem. Soc.* **2016**, *138*, 2.
- (2) (a) Ma, S.; Gu, Z. *Angew. Chem., Int. Ed.* **2005**, *44*, 7512. (b) Shi, F.; Larock, R. *Top. Curr. Chem.* **2009**, *292*, 123.
- (3) Heck, R. F. *J. Organomet. Chem.* **1972**, *37*, 389.
- (4) Tsui, G. C.; Lautens, M. In *Domino Reactions: Concepts for Efficient Organic Synthesis*; Tietze, L. F., Ed.; Wiley-VCH: Weinheim, 2014; pp 67–103.
- (5) (a) Campo, M. A.; Huang, Q.; Yao, T.; Tian, Q.; Larock, R. C. *J. Am. Chem. Soc.* **2003**, *125*, 11506. For other examples see: (b) Tian, Q.; Larock, R. C. *Org. Lett.* **2000**, *2*, 3329. (c) Larock, R. C.; Tian, Q. *J. Org. Chem.* **2001**, *66*, 7372. (d) Huang, Q.; Fazio, A.; Dai, G.; Campo, M. A.; Larock, R. C. *J. Am. Chem. Soc.* **2004**, *126*, 7460. (e) Zhao, J.; Larock, R. C. *J. Org. Chem.* **2006**, *71*, 5340. (f) Shintani, R.; Otomo, H.; Ota, K.; Hayashi, T. *J. Am. Chem. Soc.* **2012**, *134*,

7305. (g) Wang, M.; Zhang, X.; Zhuang, Y.-X.; Xu, Y.-H.; Loh, T.-P. *J. Am. Chem. Soc.* **2015**, *137*, 1341.

- (6) Huang, Q.; Larock, R. C. *Tetrahedron Lett.* **2009**, *50*, 7235.
- (7) Piou, T.; Bunescu, A.; Wang, Q.; Neuville, L.; Zhu, J. *Angew. Chem., Int. Ed.* **2013**, *52*, 12385.
- (8) Baudoin, O. *Acc. Chem. Res.* **2017**, *50*, 1114.
- (9) Holstein, P.; Dailler, D.; Vantourout, J.; Shaya, J.; Millet, A.; Baudoin, O. *Angew. Chem., Int. Ed.* **2016**, *55*, 2805.
- (10) (a) Lapointe, D.; Fagnou, K. *Chem. Lett.* **2010**, *39*, 1118. (b) Ackermann, L. *Chem. Rev.* **2011**, *111*, 1315.
- (11) (a) Baudoin, O.; Herrbach, A.; Guéritte, F. *Angew. Chem., Int. Ed.* **2003**, *42*, 5736. (b) Barder, T. E.; Walker, S. D.; Martinelli, J. R.; Buchwald, S. L. *J. Am. Chem. Soc.* **2005**, *127*, 4685.
- (12) (a) Chaumontet, M.; Piccardi, R.; Audic, N.; Hitce, J.; Peglion, J.-L.; Clot, E.; Baudoin, O. *J. Am. Chem. Soc.* **2008**, *130*, 15157. (b) Kefalidis, C. E.; Davi, M.; Holstein, P. M.; Clot, E.; Baudoin, O. *J. Org. Chem.* **2014**, *79*, 11903.
- (13) Hu, T.-J.; Zhang, G.; Chen, Y.-H.; Feng, C.-G.; Lin, G.-Q. *J. Am. Chem. Soc.* **2016**, *138*, 2897.
- (14) The 1,4-Pd shift was initially proposed to occur through other types of mechanisms including oxidative addition/reductive elimination and concerted H transfer:^{2b} (a) Singh, A.; Sharp, P. R. *J. Am. Chem. Soc.* **2006**, *128*, 5998. (b) Mota, A. J.; Dedieu, A.; Bour, C.; Suffert, J. *J. Am. Chem. Soc.* **2005**, *127*, 7171. (c) Mota, A. J.; Dedieu, A. *J. Org. Chem.* **2007**, *72*, 9669.
- (15) Pedroni, J.; Cramer, N. *Angew. Chem., Int. Ed.* **2015**, *54*, 11826.
- (16) (a) Guyonnet, M.; Baudoin, O. *Org. Lett.* **2012**, *14*, 398. (b) Rocaboy, R.; Dailler, D.; Baudoin, O. *Org. Lett.* **2018**, *20*, 772. (c) Rocaboy, R.; Dailler, D.; Zellweger, F.; Neuburger, M.; Salomé, C.; Clot, E.; Baudoin, O. *Angew. Chem., Int. Ed.* **2018**, *57*, 12131.
- (17) Dailler, D.; Rocaboy, R.; Baudoin, O. *Angew. Chem., Int. Ed.* **2017**, *56*, 7218.
- (18) Delong, W.; Lanying, W.; Yongling, W.; Shuang, S.; Juntao, F.; Xing, Z. *Eur. J. Med. Chem.* **2017**, *130*, 286.
- (19) (a) Robertson, J.; Stevens, K. *Nat. Prod. Rep.* **2014**, *31*, 1721. (b) Pilli, R. A.; Rosso, G. B.; de Oliveira, M. C. F. *Nat. Prod. Rep.* **2010**, *27*, 1908.
- (20) Grote, R.; Zeeck, A.; Stümpfel, J.; Zähler, H. *Liebigs Ann. Chem.* **1990**, *1990*, 525.
- (21) Aoyagi, Y.; Manabe, T.; Ohta, A.; Kurihara, T.; Pang, G.-L.; Yuhara, T. *Tetrahedron* **1996**, *52*, 869.
- (22) Nonaromatic α,β -unsaturated ketone reactants failed to give the corresponding arylidene pentanones.
- (23) Park, S.; Kadayat, T. M.; Jun, K.-Y.; Magar, T. B. T.; Bist, G.; Shrestha, A.; Lee, E. S.; Kwon, Y. *Eur. J. Med. Chem.* **2017**, *125*, 14.
- (24) Menezes, J. C. J. M. D. S. *RSC Adv.* **2017**, *7*, 9357.