

Millikelvin Reactive Collisions between Sympathetically Cooled Molecular Ions and Laser-Cooled Atoms in an Ion-Atom Hybrid Trap

Felix H. J. Hall and Stefan Willitsch*

Department of Chemistry, University of Basel, Klingelbergstrasse 80, 4056 Basel, Switzerland

(Received 29 August 2012; published 5 December 2012)

We report on a study of cold reactive collisions between sympathetically cooled molecular ions and laser-cooled atoms in an ion-atom hybrid trap. Chemical reactions were studied at average collision energies $\langle E_{\text{coll}} \rangle / k_B \geq 20$ mK, about 2 orders of magnitude lower than has been achieved in previous experiments with molecular ions. Choosing $\text{N}_2^+ + \text{Rb}$ as a prototypical system, we find that the reaction rate is independent of the collision energy within the range studied, but strongly dependent on the internal state of Rb. Highly efficient charge exchange four times faster than the Langevin rate was observed with Rb in the excited $(5p) \ ^2P_{3/2}$ state. This observation is rationalized by a capture process dominated by the charge-quadrupole interaction and a near resonance between the entrance and exit channels of the system. Our results provide a test of classical models for reactions of molecular ions at the lowest energies reached thus far.

DOI: [10.1103/PhysRevLett.109.233202](https://doi.org/10.1103/PhysRevLett.109.233202)

PACS numbers: 34.50.-s, 34.70.+e, 37.10.-x

The combination of radio frequency (rf) ion traps with magneto-optical or optical-dipole traps for the simultaneous confinement of cold ions and atoms has recently enabled the study of ion-neutral interactions at temperatures as low as a few millikelvin [1–7]. Hybrid trapping techniques have opened up perspectives for the exploration of new mesoscopic quantum systems [8], for quantum interfaces between atoms and ions [9], and for new methods to cool ions to ultralow temperatures [6,10]. Moreover, chemical experiments with ions [4,5,7,11] have started to approach an energy regime so far restricted to neutrals [12–15] in which the quantum character of the collision can strongly influence the chemical dynamics [16–18].

Until now, studies of reactive processes in hybrid traps have been restricted to atomic ions and neutral atoms. Despite the apparent simplicity of these collision systems, a diverse range of chemical phenomena has been observed. The effects range from the radiative formation of molecules and charge exchange in $\text{Ca}^+ + \text{Rb}$ [4], the occurrence of unusually fast reaction rates in $\text{Yb}^+ + \text{Ca}$ [5] and the influence of the ion's hyperfine state on the reactivity in $\text{Yb}^+ + \text{Rb}$ [7]. However, experiments with atomic species can only cover a fraction of the wealth of chemical phenomena, and it is necessary to extend studies to molecules to be able to fully explore the diversity of reactive effects at ultralow energies.

Here, we extend ion-neutral hybrid-trapping techniques to the study of cold reactive collisions with molecular species by sympathetically cooling molecular ions by their interaction with laser-cooled atomic ions [19] and immersing them in a cloud of ultracold atoms in a magneto-optical trap. To our knowledge, this is the first time that chemical reactions with molecular ions have been studied at collision energies down to $\langle E_{\text{col}} \rangle / k_B \approx 20$ mK, about 2 orders of magnitude lower than has been achieved in previous experiments [20–22].

By choosing sympathetically cooled N_2^+ ions and laser-cooled Rb atoms as a prototypical system, we characterize important features of cold reactive collisions between molecular ions and alkali atoms. We observe charge exchange and a strong dependence of the reaction rate on the electronic state of Rb. Whereas reactions with Rb in the $(5p) \ ^2P_{3/2}$ state occur fast, consistent with a classical capture model, reactions with ground state atoms were found to be at least 2 orders of magnitude slower. The fast excited state rate is explained by a long range capture of the atom by the molecular ion dominated by the charge-quadrupole interaction and a near resonance between the entrance and exit channels at short range. Our results highlight the importance of resonant effects in cold ion-neutral collisions and provide a test of classical reaction models for molecular ions at the lowest energies reached thus far.

Our hybrid-trapping apparatus has already been described in Ref. [4]. Here, N_2^+ ions were generated in a linear rf ion trap from room temperature background nitrogen gas using [2 + 1] resonance-enhanced multiphoton-ionization (REMPI) via the Q branch of the vibrationless $X^1\Sigma_g^+ \rightarrow a''^1\Sigma_g^+$ transition [23]. The ions were subsequently thermalized through collisions with the background gas [23].

The N_2^+ ions were sympathetically cooled by the interaction with laser-cooled Ca^+ ions to form Coulomb crystals [19]. Images of the crystals were obtained by collecting the spatially resolved laser-cooling fluorescence of the Ca^+ ions using a CCD camera coupled to a microscope. The number of ions in the crystals were determined by comparisons of the experimental images with molecular dynamics (MD) simulations, see Refs. [24,25].

After ion loading, the Coulomb crystals were translated along the axis of the rf trap using a static electric field to be overlapped with the atoms. Ultracold Rb atoms at densities $n_{\text{Rb}} = 4 \times 10^8 \text{ cm}^{-3}$ were generated by laser-cooling in the

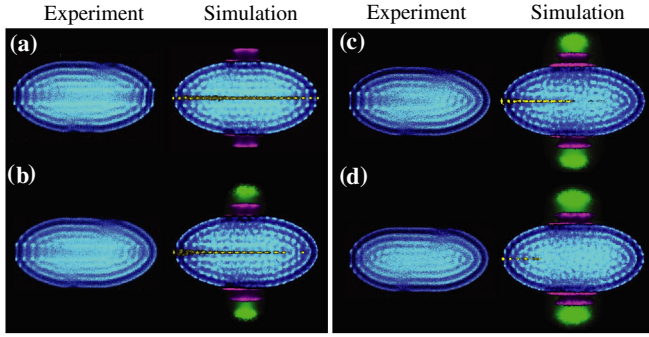


FIG. 1 (color online). Experimental false-color laser-cooling-fluorescence images of Coulomb crystals recorded over the course of reaction with ultracold Rb atoms. Sympathetically cooled ions are made visible in molecular dynamics simulations. Color code: blue: laser-cooled Ca^+ ions; yellow: N_2^+ reactant ions; green: Rb^+ product ions; magenta: CaH^+ , CaOH^+ . See text for details.

magneto-optical trap [4]. Rate constants for reactive collisions between ions and atoms were determined from the decrease of the number $N_{\text{N}_2^+}$ of N_2^+ reactant ions over the reaction time t and fitting the data to an integrated pseudofirst-order rate law $\ln[N_{\text{N}_2^+}(t)/N_{\text{N}_2^+}(t=0)] = -k_{\text{pfo}}t$. Second-order rate constants k were calculated from the pseudofirst-order constants k_{pfo} using $k = k_{\text{pfo}}/n_{\text{Rb}}$. Both the N_2^+ and Ca^+ ions react with Rb. However, the total rate for $\text{Ca}^+ + \text{Rb}$ is 2 orders of magnitude smaller [4] than for $\text{N}_2^+ + \text{Rb}$ so that the reorganization of the Coulomb crystals caused by loss of Ca^+ could be neglected over the time scale of reactions with N_2^+ .

Figures 1(a)–1(d) show a sequence of Coulomb crystal images and their MD simulations recorded during a typical reaction experiment. The fluorescence of the Rb atoms was blocked by a color filter so that only the Ca^+ fluorescence (blue) is visible. The N_2^+ ions can only indirectly be seen as a dark region in the center of the crystals, but have been made visible in yellow in the simulations. Initially, Ca^+ Coulomb crystals containing ≈ 1000 ions were loaded with typically ≈ 25 N_2^+ ions [Fig. 1(a)]. During the experiment, the N_2^+ ions were removed from the crystals by charge-exchange collisions with Rb resulting in a reduction of the dark crystal core [Figs. 1(b)–1(d)]. The Rb^+ product ions (green in the simulations) remained trapped and were sympathetically cooled by the Ca^+ ions. The magenta-colored ions in the simulated images represent CaH^+ and CaOH^+ , which are impurities from reactions of Ca^+ with background H_2 and H_2O gas, respectively, occurring during the loading period.

To prove the identity of the reaction products, resonant-excitation mass spectra were recorded [26]. Briefly, an additional rf drive voltage applied to the trap electrodes was used to resonantly excite the radial motion of specific ion species. The motion of the excited ions couples to the laser-cooled Ca^+ ions and leads to a modulation of their laser-cooling fluorescence around the resonance frequencies. Figure 2

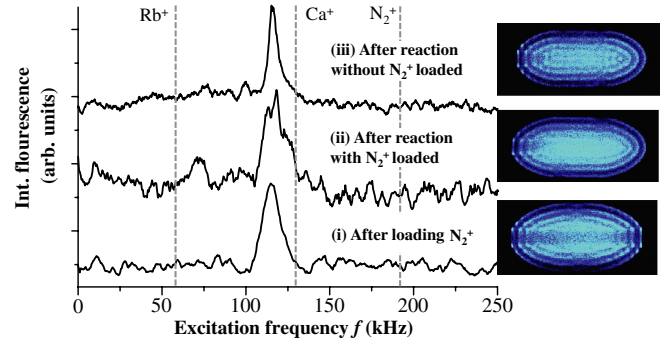


FIG. 2 (color online). Resonant-excitation mass spectra of Coulomb crystals (insets) loaded with sympathetically cooled N_2^+ ions (i) before and (ii) after reaction with ultracold Rb atoms. The dashed lines indicate the single-ion resonance frequencies of the relevant species. The weak feature around 70 kHz in (ii) corresponds to Rb^+ product ions from charge transfer of N_2^+ with Rb. (iii) Spectrum of a crystal prepared without N_2^+ ions recorded after the same reaction time as in (ii).

shows mass spectra of Coulomb crystals (i) before and (ii) after reaction. Additionally, a pure Ca^+ crystal (iii) was reacted for the same duration as in (ii) to illustrate the effect of the slow background reaction $\text{Ca}^+ + \text{Rb}$. The insets show images of the corresponding crystals and the vertical lines indicate the theoretical single-ion resonance frequencies of the relevant species.

Whereas the spectrum recorded before reaction only shows a single resonance corresponding to the excitation of the Ca^+ ions, the spectrum after reaction in (ii) shows an additional weak feature around 70 kHz assigned to sympathetically cooled Rb^+ product ions. Note that the positions of the resonances in a Coulomb crystal are slightly shifted from the single-ion frequencies because of Coulomb-coupling effects [26]. Resonances for the CaOH^+ , CaH^+ , and the reactant N_2^+ ions, as well as for the Rb^+ ions formed by the background reaction $\text{Ca}^+ + \text{Rb}$ in (iii) could not be observed in the spectra under the present conditions. MD simulations revealed that their numbers were too small to exert any noticeable back action on the Ca^+ fluorescence at the excitation amplitudes used. Higher amplitudes would have led to the melting of the crystals around the Ca^+ resonance and were therefore precluded.

Figure 3(a) shows the dependence of the rate constant on the collision energy. In the experiments, the Rb kinetic energies ($\langle E_{\text{kin}} \rangle / k_B \approx 200 \mu\text{K}$) were much smaller than the ion energies (≥ 20 mK) so that the collision energies were dominated by the contribution of the ions. Their motion can be separated into two different components, a slow secular motion in the time-averaged pseudopotential of the trap and a fast micromotion driven by the rf fields [19]. The secular ($\langle E_{\text{sec}} \rangle / k_B = 12$ mK for Ca^+ and 14 mK for N_2^+) and micromotion energies were determined from the ion trajectories obtained in the MD simulations of the crystal images [24]. The total kinetic energy E_{tot} of the ions was governed by their micromotion energy which vanishes for

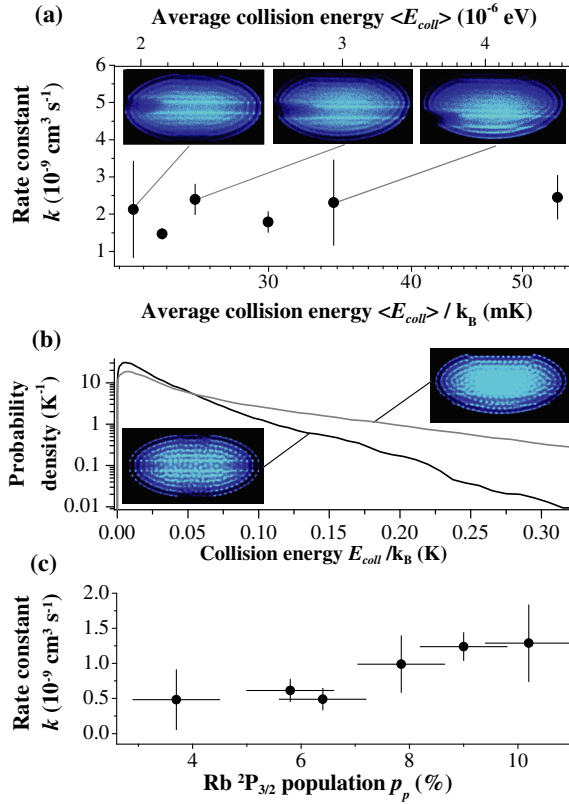


FIG. 3 (color online). (a) Rate constants for $\text{N}_2^+ + \text{Rb}$ as a function of the collision energy. Each data point represents an average over three independent measurements, the error bars indicate the statistical uncertainty (2σ). Insets: Coulomb crystal images for the three data points indicated. (b) $\text{N}_2^+ + \text{Rb}$ collision-energy distributions from MD simulations for the data points with the lowest (black) and highest energies (grey) in (a). Insets: simulated images of the corresponding crystals. (c) Rate constant as a function of the population in the Rb ($5p$) $^2P_{3/2}$ state.

ions located on the central trap axis and increases quadratically with the distance [19]. The lowest achievable energies are limited by the precision with which the N_2^+ ions can be localized on the axis. Our present axialization procedures (to be described in detail in Ref. [27]) allowed for a micro-motion energy limit corresponding to 17 mK.

For the measurements in Fig. 3(a), Coulomb crystals with strings of nitrogen ions [like the ones shown in Fig. 1(a)] were aligned parallel to the trap axis where all ions exhibit approximately the same micromotion energy [24]. The collision energies were varied by displacing the Coulomb crystals from the axis using static electric fields causing the N_2^+ ions to move into regions where they acquire an increased micromotion energy (see insets). The dislocated crystals exhibit an asymmetric ion distribution because the heavier Ca^+ ions are more strongly displaced than the lighter N_2^+ ions as a consequence of the mass dependence of the effective trapping potential [19]. The collision-energy distributions for the data points with the lowest and highest

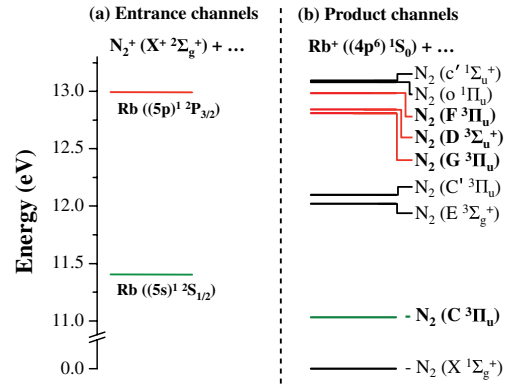


FIG. 4 (color online). Total electronic energies of (a) $\text{N}_2^+ + \text{Rb}$ reactant (entrance) and (b) $\text{Rb}^+ + \text{N}_2$ product channels. The channels are characterized by the electronic states of the species as indicated. Charge transfer is most efficient in case of a near-resonance between the entrance and product channels. The product channels closest in energy to the $\text{N}_2^+ + \text{Rb}$ ($5s$) and ($5p$) entrance channels are indicated in green and red, respectively. See text for details.

energies in Fig. 3(a) obtained from the MD simulations are displayed in Fig. 3(b) alongside the corresponding simulated images. Within the uncertainty limits, the rate constant was observed to be essentially independent of the collision energy in the interval studied.

Because the ultracold atoms are constantly excited during laser cooling, a fraction of the reactions occurs with Rb atoms in the ($5p$) $^2P_{3/2}$ state. The population in the excited state was adjusted by varying the intensity of the Rb cooling laser [4]. The rate constant shows an approximately linear increase with the ($5p$) $^2P_{3/2}$ population [Fig. 3(c)]. Following Ref. [4], a kinetic model $k = \frac{1}{2}[k_s(1 + p_s) + k_p p_p]$, where $k_{s,p}$ and $p_{s,p}$ denote the rate constants and populations in the Rb ($5s$, $5p$) states, respectively, was fitted to the data. The fit yielded $k_p = 2.4(13) \times 10^{-8} \text{ cm}^3 \text{ s}^{-1}$ and an upper bound $k_s \leq 2 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$. The uncertainty of k_p is dominated by a systematic error of $\approx 50\%$ in the Rb density.

Because of the large difference between the ionization energies of N_2 and Rb, a minimum of 11.4 eV of energy is released in the reaction. By comparison, the rotational-vibrational energy of the N_2^+ ions (≈ 0.025 eV for a room temperature distribution) as well as the collision energy ($\approx 10^{-6}$ eV) is small, so that the available energy is entirely dominated by the electronic contribution. Figure 4 shows an energy diagram depicting the relevant electronic entrance and product channels (see Ref. [28] for an overview of the electronic states of N_2).

Previous studies using keV ion beams showed that charge exchange between alkali atoms and N_2^+ is most efficient in a near-resonant transfer of the electron into a Rydberg state of the neutral product molecule which is built on an ion core with the same electronic configuration as the reactant ion [29]. In the present case, the $\text{N}_2^+(X^+) + \text{Rb}(5s)$ entrance channel is closest in energy to a product

channel forming N_2 in the $C^3\Pi_u$ electronic state (energy mismatch $\Delta E = 372$ meV, see Fig. 4). However, electron capture by N_2^+ in its $X^+2\Sigma_g^+$ electronic ground state [molecular orbital configuration $(2\sigma_g)^2(2\sigma_u)^2(1\pi_u)^4 \times (3\sigma_g)^1$] to form N_2 in the C state [$(2\sigma_g)^2(2\sigma_u)^1(1\pi_u)^4 \times (3\sigma_g)^2(1\pi_g)^1$] entails a significant rearrangement of the electronic configuration. This process is, therefore, expected to be inefficient, in line with the observation of a slow reaction rate in this channel. On the other hand, the excited $N_2^+(X^+) + Rb(5p)$ entrance channel is near resonant with product channels forming N_2 in the the close-lying $G^3\Pi_u$, $D^3\Sigma_u^+$ and $F^3\Pi_u$ states (energy mismatches $\Delta E = 183$ meV, 151 meV, and 8 meV). These states are among the lowest Rydberg states of N_2 built on the $N_2^+ X^+ 2\Sigma_g^+$ (in case of the G , D states) and $A^+ 2\Pi_u$ (in case of the F state) ion cores. The F state is heavily mixed with the G state and therefore also contains X^+ core character [30]. Thus, capture of the electron into all three states is expected to be efficient, in agreement with the fast rate observed in the excited channel. Moreover, the increased density of states in this region (see Fig. 4) and their short lifetimes (and therefore large resonance widths) [30] also contribute to the enhancement of the reaction probability. The Rydberg molecules subsequently predissociate into the atomic fragments $N(^4S) + N(^2D)$ [29] which, however, cannot be detected in the present experiment.

In previous studies on cold ion-neutral collisions in hybrid traps [1,7], the rates were found to be in agreement with a limiting value set by the classical Langevin theory which only takes into account the interaction between the charge of the ion and the induced dipole of the neutral [31]. The rate constant for the excited channel determined above [$k_p = 2.4(13) \times 10^{-8} \text{ cm}^3 \text{ s}^{-1}$], however, is about four times larger than the value for the Langevin collision rate constant [$k^{(L)} = 6.6 \times 10^{-9} \text{ cm}^3 \text{ s}^{-1}$]. This finding indicates that additional long range intermolecular forces must be effective. The most likely candidate is the interaction of the charge of the ion with the permanent electric quadrupole moment of Rb generated by the anisotropic charge distribution in the $(5p)^2P_{3/2}$ state [32].

To quantify this effect, the rate constant was calculated with a classical model based on an interaction potential [33] incorporating the Langevin and charge-quadrupole interactions. The quadrupole moment of Rb $(5p)^2P_{3/2}$ was computed within the single-electron approximation following Ref. [34]. A capture approximation [35] was adopted; i.e., the reaction was assumed to be governed by the long range interactions and to occur with unit probability once the reaction complex was formed. The classical reaction cross section $\sigma^{(c)} = \pi b^2$ was calculated from the maximum possible impact parameter b for which the height of the barrier in the centrifugally corrected potential does not exceed the collision energy [36]. From $\sigma^{(c)}$, the corresponding rate constant at $E_{\text{coll}}/k_B = 23$ mK was

obtained to be $k_p^{(c)} = 1.7 \times 10^{-8} \text{ cm}^3 \text{ s}^{-1}$, which agrees with the experimental value within its uncertainty limits. The agreement supports the conclusion that the reaction rate is close to the collision rate and that the short range charge-transfer probability is near unity, in line with the mechanism discussed above. Note that the charge-quadrupole interaction leads to a rate constant weakly dependent on the energy ($k \propto E_{\text{coll}}^{-1/6}$) which, however, was not observable over the limited energy interval studied in Fig. 3(a) within the experimental error boundaries.

Notwithstanding the approximations inherent in the model, the result underlines the necessity to reach beyond the Langevin picture to explain the dynamics in the present system. It also suggests that—at least for the present case—a classical capture model seems to adequately reproduce the kinetics in the millikelvin regime. This conclusion is consistent with the theoretical results of Refs. [18,35] which predict a robust performance of these models down to ultracold ($< \text{mK}$) temperatures for all but the lightest systems. This behavior was rationalized by a cancellation of tunneling and reflection effects at the centrifugal barrier [35].

In conclusion, we have studied reactions between sympathetically cooled N_2^+ ions and ultracold Rb atoms at average collision energies down to ≈ 20 mK in an ion-atom hybrid trap. Our results highlight the importance of resonant electronic effects in electron-transfer processes between ultracold alkali atoms and cold molecular ions. The opportunities for resonant charge transfer are markedly enhanced by the dense structure of rotational-vibrational levels [29,30], a unique property of molecular systems compared to the atomic species previously studied in hybrid traps [4,5,7]. A classical capture model was found to adequately describe the kinetics down to the lowest collision energies in cases in which the short range electron-transfer probability is close to unity. Short range effects, such as the rotational-vibrational state of the molecular ion which are not accounted for in the capture picture, could, in principle, modify the reaction probabilities. Because the present experiments were performed with rotationally thermalized ions, we plan further experiments with rotationally and vibrationally state-selected nitrogen ions to characterize the possible role of these effects [25].

We acknowledge support from the Swiss National Science Foundation (Grant No. PP0022_118921), the University of Basel, and the COST Action MP1001. We thank Dr. Nuria Plattner for helpful discussions.

*stefan.willitsch@unibas.ch

- [1] A. T. Grier, M. Cetina, F. Oručević, and V. Vuletić, *Phys. Rev. Lett.* **102**, 223201 (2009).
- [2] C. Zipkes, S. Palzer, C. Sias, and M. Köhl, *Nature (London)* **464**, 388 (2010).
- [3] S. Schmid, A. Härter, and J. H. Denschlag, *Phys. Rev. Lett.* **105**, 133202 (2010).

- [4] F.H.J. Hall, M. Aymar, N. Bouloufa-Maafa, O. Dulieu, and S. Willitsch, *Phys. Rev. Lett.* **107**, 243202 (2011).
- [5] W.G. Rellergert, S.T. Sullivan, S. Kotochigova, A. Petrov, K. Chen, S.J. Schowalter, and E.R. Hudson, *Phys. Rev. Lett.* **107**, 243201 (2011).
- [6] K. Ravi, S. Lee, A. Sharma, G. Werth, and S.A. Rangwala, *Nat. Commun.* **3**, 1126 (2012).
- [7] L. Ratschbacher, C. Zipkes, C. Sias, and M. Köhl, *Nat. Phys.* **8**, 649 (2012).
- [8] R. Côté, V. Kharchenko, and M.D. Lukin, *Phys. Rev. Lett.* **89**, 093001 (2002).
- [9] Z. Idziaszek, T. Calarco, and P. Zoller, *Phys. Rev. A* **76**, 033409 (2007).
- [10] E.R. Hudson, *Phys. Rev. A* **79**, 032716 (2009).
- [11] C. Zipkes, S. Palzer, L. Ratschbacher, C. Sias, and M. Köhl, *Phys. Rev. Lett.* **105**, 133201 (2010).
- [12] J. Weiner, V.S. Bagnato, S. Zilio, and P.S. Julienne, *Rev. Mod. Phys.* **71**, 1 (1999).
- [13] S. Ospelkaus, K.-K. Ni, D. Wang, M.H.G. de Miranda, B. Neyenhuis, G. Quéméner, P.S. Julienne, J.L. Bohn, D.S. Jin, and J. Ye, *Science* **327**, 853 (2010).
- [14] K.-K. Ni, S. Ospelkaus, D. Wang, G. Quéméner, B. Neyenhuis, M.H.G. de Miranda, J.L. Bohn, J. Ye, and D.S. Jin, *Nature (London)* **464**, 1324 (2010).
- [15] A.B. Henson, S. Gersten, Y. Shagam, J. Narevicius, and E. Narevicius, *Science* **338**, 234 (2012).
- [16] A.K. Belyaev, S.A. Yakovleva, M. Tacconi, and F.A. Gianturco, *Phys. Rev. A* **85**, 042716 (2012).
- [17] B. Gao, *Phys. Rev. Lett.* **104**, 213201 (2010).
- [18] B. Gao, *Phys. Rev. A* **83**, 062712 (2011).
- [19] S. Willitsch, *Int. Rev. Phys. Chem.* **31**, 175 (2012).
- [20] S. Willitsch, M.T. Bell, A.D. Gingell, S.R. Procter, and T.P. Softley, *Phys. Rev. Lett.* **100**, 043203 (2008).
- [21] M.A. Smith, *Int. Rev. Phys. Chem.* **17**, 35 (1998).
- [22] R. Otto, J. Mikosch, S. Trippel, M. Weidemüller, and R. Wester, *Phys. Rev. Lett.* **101**, 063201 (2008).
- [23] X. Tong, D. Wild, and S. Willitsch, *Phys. Rev. A* **83**, 023415 (2011).
- [24] M.T. Bell, A.D. Gingell, J. Oldham, T.P. Softley, and S. Willitsch, *Faraday Discuss. Chem. Soc.* **142**, 73 (2009).
- [25] X. Tong, A.H. Winney, and S. Willitsch, *Phys. Rev. Lett.* **105**, 143001 (2010).
- [26] B. Roth, P. Blythe, and S. Schiller, *Phys. Rev. A* **75**, 023402 (2007).
- [27] F.H.J. Hall, P. Eberle, G. Hegi, M. Raoult, O. Dulieu, and S. Willitsch (to be published).
- [28] A. Lofthus and P.H. Krupenie, *J. Phys. Chem. Ref. Data* **6**, 113 (1977).
- [29] A.B. van der Kamp, P.C. Cosby, and W.J. van der Zande, *Chem. Phys.* **184**, 319 (1994).
- [30] B.R. Lewis, A.N. Heays, S.T. Gibson, H. Lefebvre-Brion, and R. Lefebvre, *J. Chem. Phys.* **129**, 164306 (2008).
- [31] G. Gioumouzis and D.P. Stevenson, *J. Chem. Phys.* **29**, 294 (1958).
- [32] F. Pirani, G.S. Maciel, D. Cappelletti, and V. Aquilanti, *Int. Rev. Phys. Chem.* **25**, 165 (2006).
- [33] M. Krych, W. Skomorowski, F. Pawłowski, R. Moszynski, and Z. Idziaszek, *Phys. Rev. A* **83**, 032723 (2011).
- [34] I.I. Sobelman, *Atomic Spectra and Radiative Transitions* (Springer, Berlin, 1979).
- [35] E.I. Dashevskaya, A.I. Maergoiz, J. Troe, I. Litvin, and E.E. Nikitin, *J. Chem. Phys.* **118**, 7313 (2003).
- [36] R.D. Levine, *Molecular Reaction Dynamics* (Cambridge University Press, Cambridge, England, 2005).